

T.C.
DOKUZ EYLÜL UNIVERSITY
İZMİR INTERNATIONAL
BIOMEDICINE AND GENOME
INSTITUTE

**THE EFFECTS OF RUTHENIUM COMPLEXES IN
OVARIAN CANCER**

GÜLNUR ÇIRAK

MOLECULAR BIOLOGY AND GENETICS MASTER'S
PROGRAM

MASTER OF SCIENCE THESIS

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İZMİR ULUSLARARASI BİYOTİP VE GENOM ENSTİTÜSÜ



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BASBAŞI

Prof. Dr. Hülya AYAR KAYALI

Dokuz Eylül Üniversitesi İzmir Uluslararası Biyotip ve Genom Enstitüsü

ÜYELİK ÜYESİ

Prof. Dr. Gül SUBAŞI
Dokuz Eylül Üniversitesi
Anorganik Kimya Anabilim Dalı

ÜYELİK ÜYESİ

Prof. Dr. Sinan AKGÖL
Ege Üniversitesi
Biyokimya Ana Bilim Dalı

YEDEK ÜYE

Prof. Dr. Nur Arslan
Dokuz Eylül Üniversitesi
Moleküler Tıp Anabilim Dalı

YEDEK ÜYE

Prof. Dr. Sonay ŞANLIER
Ege Üniversitesi
Biyokimya Ana Bilim Dalı

TABLE OF CONTENTS

TABLE OF CONTENTS.....	i
LIST OF TABLES	iii
LIST OF FIGURES	iv
ABBREVIATIONS	vii
ACKNOWLEDGEMENTS.....	viii
ABSTRACT.....	1
ÖZET.....	2
1. INTRODUCTION AND AIM.....	3
2. GENERAL INFORMATION	4
2.1. Ovarian Cancer.....	4
2.2. Etiology of Ovarian Cancer	5
2.3. Epidemiology.....	6
2.4. The Staging of Ovarian Cancers	7
2.5. Cancer Treatment and Chemotherapy.....	9
2.5.1. <i>Platinum-based Chemotherapeutics</i>	10
2.5.1.1. <i>Carboplatin</i>	11
2.5.2. <i>Non-Platinum-based Chemotherapeutics</i>	11
2.5.2.1. <i>Paclitaxel</i>	11
2.5.2.2. <i>Ruthenium</i>	12
2.6. Metabolism in Cancer	14
2.6.1. <i>Glucose metabolism</i>	15
2.6.2. <i>Tricarboxylic Acid Cycle</i>	17
2.6.3. <i>Electron Transport Chain</i>	18
2.6.4. <i>Anaerobic Respiration</i>	19

2.6.5. <i>Amino Acid Metabolism</i>	19
2.6.6. <i>Metabolism in Ovarian Cancer</i>	20
3. MATERIALS AND METHODS	22
3.1. Type of Research	22
3.2. Time and Place of Research	22
3.3. Study Plan and Calendar	22
3.4. Materials	22
3.4.1. <i>Machines</i>	24
3.4.2. <i>Cell Culture Materials</i>	25
3.5. Methods	26
3.5.1. <i>Cell Culture Conditions</i>	26
3.5.2. <i>Cell Viability Assay</i>	26
3.5.3. <i>Preparation of Samples for HPLC Analysis</i>	26
3.5.4. <i>HPLC Conditions for Metabolites</i>	27
3.5.5. <i>Colorimetric Determination of Pyruvic Acid</i>	27
3.5.6. <i>Statistical analysis</i>	28
4. RESULTS	29
4.1. Cytotoxic Activities of Chemotherapeutic Agents on Ovarian Cell Lines	29
4.2. Intra and Extra-Glucose Levels of Ovarian Cell Lines	36
4.3. Intracellular TCA Cycle Metabolites of Ovarian Cell Lines	39
4.4. Lactic Acid Production in Ovarian Cell Lines	44
5. DISCUSSION	45
6. CONCLUSION AND RECOMMENDATIONS	50
7. REFERENCES	51
8. APPENDIX	74
8.1. Curriculum Vitae	74

LIST OF TABLES

Table 1 FIGO Staging System for Ovarian Cancer (Adapted from Prat, 2015)	8
Table 2 The chemicals used in this study.....	23
Table 3 The machines used in this study.....	24
Table 4 Used Materials in Cell Culture.....	25
Table 5 IC ₅₀ values of MTT Assay (μM: micromolar, nM: nanomolar, “-“: no data).....	35

LIST OF FIGURES

Figure 1 Anatomy of the reproductive system of women (Adapted from National Cancer Institute, 2019).....	4
Figure 2 Risk factors associated with Ovarian Cancer	6
Figure 3 Global Incidence of Ovarian Cancer in 2018 (for all ages). ASR displays an age-standardized rate (WHO, 2018).....	7
Figure 4 Stages of Ovarian Cancer (Figure created in the Mind the Graph platform, www.mindthegraph.com).	9
Figure 5 The structures of three platinum drugs that are currently in clinical use (Adapted from Wheate, Walker, Craig, & Oun, 2010).	10
Figure 6 Structures of KP1019 and NAMI-A, respectively (Adapted from Alessio, Mestroni, Bergamo, & Sava, 2004).	13
Figure 7 Images of 1b treated OVCAR-3 cells for variable incubation time. DAPI (blue) stain represents nuclei, the green fluorescence stain (green) represents complex 1b, and the overlay demonstrates the cellular relation of these compounds with DNA. (Adapted from Tavsan, Yaman, Subasi, & Ayar Kayalı, 2018).	14
Figure 8 Energy metabolism in the cell (ATP and other cofactors have been removed for simplicity. G6P: glucose-6-phosphate, PPP: pentose phosphate pathway, F6P: fructose-6-phosphate, α KG: alpha-Ketoglutarate, TCA: tricarboxylic acid, NADPH: nicotinamide adenine dinucleotide phosphate. Figure created via biorender, www.biorender.com, and adapted from DeBerardinis & Chandel, 2016).	17
Figure 9 Outline of the study.....	22
Figure 10 Ru (III) complexes. Complex 1a, 1b, and their ligand are TSC ¹ , Complex 2, and its ligand is TSC ² (Adapted from Tavsan, Yaman, Subasi, & Ayar Kayalı, 2018).....	24
Figure 11 MTT results of complex 1a treatment in ovarian cell lines (A: OSE cell line with treatment of complex 1a for 24 h and 48 h; B: A2780 cell line with treatment of complex 1a for 24	

h and 48 h; C: A2780cis cell line with treatment of complex 1a for 24 h and 48 h; D: OVCAR-3 cell line with treatment of complex 1a for 24 h and 48 h)..... 30

Figure 12 MTT results of complex 1b treatment in ovarian cell lines (A: OSE cell line with treatment of complex 1b for 24 h and 48 h; B: A2780 cell line with treatment of complex 1b for 24 h and 48 h; C: A2780cis cell line with treatment of 1b complex for 24 h and 48 h; D: OVCAR-3 cell line with treatment of complex 1b for 24 h and 48 h). 31

Figure 13 MTT results of complex 2 treatment in ovarian cell lines (A: OSE cell line with treatment of complex 2 for 24 h and 48 h; B: A2780 cell line with treatment of complex 2 for 24 h and 48 h; C: A2780cis cell line with treatment of complex 2 for 24 h and 48 h; D: OVCAR-3 cell line with treatment of complex 2 for 24 h and 48 h). 32

Figure 14 MTT results of PAX treatment in ovarian cell lines (A: OSE cell line with treatment of PAX for 24 h and 48 h; B: A2780 cell line with treatment of PAX for 24 h and 48 h; C: A2780cis cell line with treatment of PAX for 24 h and 48 h; D: OVCAR-3 cell line with treatment of PAX for 24 h and 48 h)..... 33

Figure 15 MTT results of CAR treatment in ovarian cell lines (A: OSE cell line with treatment of CAR for 24 h and 48 h; B: A2780 cell line with treatment of CAR for 24 h and 48 h; C: A2780cis cell line with treatment of CAR for 24 h and 48 h; D: OVCAR-3 cell line with treatment of CAR for 24 h and 48 h)..... 34

Figure 16 Change of extracellular glucose levels in ovarian cell lines (A: OSE cell line; B: A2780 cell line; C: A2780cis cell line; D: OVCAR-3 cell line)..... 37

Figure 17 Change of intracellular glucose levels in ovarian cell lines (A: OSE cell line; B: A2780 cell line; C: A2780cis cell line; D: OVCAR-3 cell line)..... 38

Figure 18 Change of OAA levels in ovarian cell lines (A: OSE cell line; B: A2780 cell line; C: A2780cis cell line; D: OVCAR-3 cell line)..... 39

Figure 19 Change of alpha-KG levels in ovarian cell lines (A: OSE cell line; B: A2780 cell line; C: A2780cis cell line; D: OVCAR-3 cell line)..... 40

Figure 20 Change of succinate levels in ovarian cell lines (A: OSE cell line; B: A2780 cell line; C: A2780cis cell line; D: OVCAR-3 cell line)..... 41

Figure 21 Change of fumarate levels in ovarian cell lines (A: OSE cell line; B: A2780 cell line; C: A2780cis cell line; D: OVCAR-3 cell line)..... 42

Figure 22 Change of malate levels in ovarian cell lines (A: OSE cell line; B: A2780 cell line; C: A2780cis cell line; D: OVCAR-3 cell line)..... 43

Figure 23 Change of lactate levels in ovarian cell lines (A: OSE cell line; B: A2780 cell line; C: A2780cis cell line; D: OVCAR-3 cell line)..... 44

ABBREVIATIONS

ATP.....	Adenosine 5'-triphosphate
CAR.....	Carboplatin
D-2HG.....	D-2-hydroxyglutarate
DMSO.....	Dimethyl sulfoxide
DNPH.....	2, 4-dinitrophenylhydrazine
ETC.....	Electron transport chain
FAD.....	Flavine Adenine Dinucleotide
FH.....	Fumarate hydratase
HGSC.....	High-grade serous carcinoma
HIF-1a.....	Hypoxia-inducible factor 1-alpha
HPLC.....	High-Performance Liquid Chromatography
IDH.....	Isocitrate dehydrogenase
mtDNA.....	Mitochondrial DNA
MTT.....	3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide
NADPH.....	Nicotinamide Adenosine Dinucleotide Phosphate
NAMI-A.....	New Anti-tumor Metastasis Inhibitor A
NEAA.....	Non-essential amino acid
OAA.....	Oxaloacetate
OSE.....	Ovarian surface epithelium
OXPHOS.....	Oxidative phosphorylation
PAX.....	Paclitaxel
PPP.....	Pentose phosphate pathway
SDH.....	Succinate dehydrogenase
TCA.....	Tricarboxylic acid
α KG.....	alpha-Ketoglutarate
OC.....	Ovarian Cancer

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Gülnur ÇIRAK

THE EFFECTS OF RUTHENIUM COMPLEXES IN OVARIAN CANCER

**Gülnur Çırak, Dokuz Eylül University İzmir International Biomedicine and Genome
Institute, Balçova, 35340, İzmir/TURKEY**

ABSTRACT

Ovarian cancer has a high death incidence among females worldwide. Although first line chemotherapy with platinum/taxane combination has been used successfully in ovarian cancer, the majority of the patients relapsed and gained chemotherapy resistance. Lately, ruthenium complexes have emerged as encouraging chemotherapeutic agents to replace the platinum-based chemotherapy regarding their selective mechanisms of action along with lower toxicity, and the potential to overcome chemotherapy resistance.

It is accepted that cancer is a heterogeneous disease, and it also exhibits characteristic variations in metabolic phenotype from one cell to another. Because cancer cells repeatedly reprogram to adapt to challenging environmental changes.

In this thesis, we aimed to examine the effects of chemotherapeutic agents on ovarian cell lines. Therefore, we evaluated the alterations in the levels of glycolysis metabolites such as glucose, pyruvate; critical intermediates of TCA cycle such as OAA, α KG, succinate, fumarate, malate, and the metabolite of anaerobic respiration, lactate, with respect to the Ru, PAX, CAR concentrations and incubation time. The results proved that each ovarian cell line displayed distinct metabolic phenotypes of energy, and thus, each ovarian cell line responded in a unique way to chemotherapeutic agents. Additionally, ruthenium complexes showed better anti-tumor activity than carboplatin and lower toxicity against healthy cells than paclitaxel. Among ruthenium agents, complex 1a demonstrated the most promising results against cisplatin-resistant ovarian cancer cells.

Key Words: Ruthenium, Ovarian cancer, TCA cycle

RUTENYUM KOMPLEKSLERİNİN OVARYUM KANSERİ ÜZERİNDEKİ ETKİSİ

**Gülnur Çırak, Dokuz Eylül Üniversitesi İzmir Uluslararası Biyotıp ve Genom Enstitüsü,
Balçova, 35340, İzmir/TÜRKİYE**

ÖZET

Ovaryum kanseri, Dünya çapındaki en ölümcül kanserlerden biridir. Ovaryum kanseri tedavisinde platin/taksol kombinasyonu her ne kadar ilk aşamada başarılı olsada, hastaların büyük bir kısmı kemoterapiye direnç kazanarak, kanseri nüks etmiştir. Son yıllarda rutenyum kompleksleri gösterdiği seçici anti-kanser aktivitesi, daha az toksik olması ve dirençli hücreleri öldürebilme potansiyeli ile platin-bazlı bileşenlerin yerini alabilecek umut vadeden kemoterapi ajanları olarak ortaya çıkmıştır.

Kanserin heterojen yapıda bir hastalık olduğu kabul edilmiş bir tanımdır. Her kanser hücresi, kendisine özgü bir metabolik fenotip gösterir. Çünkü kanser hücreleri, sürekli olarak metabolizmalarını çevresel etkenlere göre programlayarak adaptasyon sağlarlar.

Bu çalışmada amacımız, kemoterapik ajanların ovaryum hücreleri üzerindeki etkilerini gözlemlemektir. Bu nedenle belli konsantrasyonlarda ve zaman aralıklarında rutenyum, paklitaksel ve karboplatin uygulanmış ovaryum hücrelerindeki glikoliz metabolitlerinden glukoz ve piruvat; Krebs döngüsünün metabolitlerinden oksaloasetat, fumarat, malat, süksinat ve alfa-ketoglutarat; oksijensiz solunum metaboliti olan laktat seviyeleri ölçüldü. Elde ettiğimiz sonuçlar, her ovaryum hücre hattının benzersiz bir enerji metabolizmasının olduğunu ve bu yüzden de her hücre hattının kemoterapiye özgün yanıtlar verdiğini gösterdi. Ek olarak, rutenyum komplekslerinin karboplatinden daha iyi bir anti-kanser ilacı olduğu ve paklitaksele kıyasla sağlıklı hücrelere daha az toksik olduğunu gösterdi. Uygulanmış rutenyumlar arasında sisplatin dirençli ovaryum hücrelerine karşı en umut vadeden sonuçları veren kompleks 1a oldu.

Anahtar Sözcükler: Ovaryum kanseri, Rutenyum, Krebs döngüsü

1. INTRODUCTION AND AIM

Cancer caused an estimated 9.6 million deaths in 2018 and still is the second major reason for death in the world (World Health Organization, 2018). Ovarian cancer (OC) is the fifth major cause of death associated with cancer among females in the world. Moreover, nearly 1.3% of females will be diagnosed with OC someday in their life, according to 2014-2016 data (WHO, 2018). Its 5-year relative survival rate is 47% due to late stage (such as stage III or IV) diagnose (American Cancer Society, 2019).

There is consequential proof showing that metal-based chemotherapeutics are encouraging opponents for cancer treatment. Platinum-derived complexes, such as cisplatin, and carboplatin, have been used on numerous kinds of cancers such as ovary, colon, and stomach (Jung & Lippard 2007; Todd & Lippard, 2009). Even though OC patients were responded mostly fine to the first line chemotherapy with platinum/taxane (around 80-90%), the most significant part of patients relapsed and developed chemotherapy resistance (Herzog, 2004; Banno et al., 2014; Stronach et al., 2015). For that reason, designing chemotherapeutic compounds with better efficiency to conquer the drug resistance in OC is a critical and urgent need.

Ruthenium (Ru) complexes are recognized as an alternative to platinum-based compounds, due to their attributes: performing various mechanisms of action, and the potential to conquer the drug resistance, along with lower toxicity. Studies showed that Ruthenium compounds are inhibitors for cell growth in numerous cancer types, such as ovarian, breast, and melanoma (Fernandez et al., 2004; Ang & Dyson, 2006; Bugarcic et al., 2009; Betanzos- Lara et al., 2012).

It is clear now that tumor cells can regulate cellular metabolism under certain circumstances. There is a lack of fundamental researches on the metabolic response of cancer cells to chemotherapeutic agents. Studying the energy metabolism of cancer cells may lead to essential evaluations that would benefit for further treatment applications. In order to evaluate the metabolic response of ovarian cells to ruthenium complexes, the study was examined the alterations of levels of glycolysis metabolites such as glucose, pyruvate; intermediates of tricarboxylic acid (TCA) cycle such as oxaloacetate (OAA), α -ketoglutarate (α KG), malate, fumarate, succinate, and the

metabolite of anaerobic respiration, lactate, with respect to ruthenium (Ru), paclitaxel (PAX), carboplatin (CAR) concentrations and incubation time.

The thesis aims to evaluate the cytotoxicity of certain chemotherapeutic agents and elucidate the effects of these agents on the energy metabolism of ovarian cancer.

2. GENERAL INFORMATION

2.1. Ovarian Cancer

As can be seen in Figure 1, the ovaries are a pair of oval-shaped glands that are located on either side of the uterus. They generate eggs that pass through the fallopian tubes into the uterus. They are also responsible for the production of hormones, such as progesterone and estrogen, which perform an essential part in the menstrual period and pregnancy (National Cancer Institute, 2019).

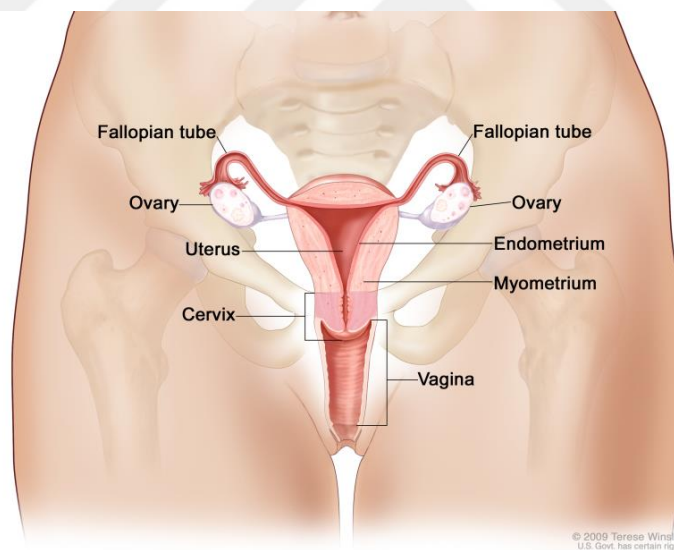


Figure 1 Anatomy of the reproductive system of women (Adapted from National Cancer Institute, 2019).

Ovarian tumors are categorized into three primary division by the anatomic structures from which they originate (Chen et al., 2003):

- The cells of the external surface of the ovary create epithelial-stromal tumors,
- The cells that generate the eggs create germ cell tumors, and
- Connective tissue cells create sex cord-stromal tumors.

Epithelial ovarian cancer is the most common type (60%) among gynecological malignancies (Siegel, Miller, & Jemal, 2018). They are classified as type I and II, based on their unique clinicopathologic and genetic aspects (Kurman & Shih, 2010). Type I exhibit somatic mutations of BRAF, KRAS, or ERBB2 genes, lack of TP53 mutations. They are comprising of low-grade endometrioid, low-grade serous, mucinous carcinomas, and clear cell. They have a good prognosis, although only less than 30% are in this group. On the other hand, tumors in type II are more aggressive and common such as high-grade serous carcinoma (HGSC) (70%). These tumors exhibit repeated mutations in oncogenes, including TP53. Non-epithelial ovarian malignancies are rarer such as sex cord-stromal malignancy. Transitional cell carcinoma was first proposed as a subtype of Brenner tumor, lacked benign factors of Brenner tumors but contained urothelial factors. These tumors are now recognized as a morphologic version of HGSC, and Brenner tumors are recognized as low-grade carcinomas (Brammer et al., 1990; Chen et al., 2003; Prat, 2012; Mutch & Prat, 2014).

2.2. Etiology of Ovarian Cancer

Ovarian cancer has numerous causes, but the most critical risk factor is having a family history of breast or ovarian cancer with BRCA1 or BRCA2 mutations, which accounts for 10% to 15% of all (Hennessy, Coleman, & Markman, 2009). The risk factors for OC summarized in Figure 2. The risk of acquiring OC increases significantly between the ages of 50 and 80 years (Glance, 2009). The studies have shown that the women who had experienced a full-term pregnancy were showed a decreased risk for developing OC compared to women who have not and the risk was minimized with each additional pregnancy (Whittemore, Harris, & Itnyre, 1992; Kvåle, Heuch, Nilssen, & Beral, 1988). While infertility looks as if enhancing the risk of OC in various studies (McGowan et al., 1988; Hartge et al., 1989; Risch, Marrett, & Howe, 1994; Mosgaard et al., 1997), however, there are some others to disagree with it (Kvåle, Heuch, Nilssen, & Beral, 1988; Franceschi et al., 1994). Moreover, some studies were reported a moderate decrease in OC risk among women who breastfed, and the risk decreases more if it is for a more extended period (Wentzensen et al., 2016; Gaitskell et al., 2017). Another study reported that oral contraceptives display a protective effect against OC with the mutations of BRCA1 and BRCA2. A protective effect was also observed for breastfeeding in patients with the mutations of BRCA1 (McLaughlin

et al., 2007). Low parity, late menopause, early menarche, and hormone replacement therapy (HRT) (especially for long-period use) increases OC risk (Lacey et al., 2002; Riman et al., 2002; Glance, 2009). Another study discovered that if all other risk elements had stayed constant, an increase in height and weight would be linked to a three percent acceleration in ovarian cancer frequency for every decade (Beral et al., 2012). In the general population, OC incidence was higher in the whites compared to other ethnic groups (Morris, Sands, & Smith, 2010).

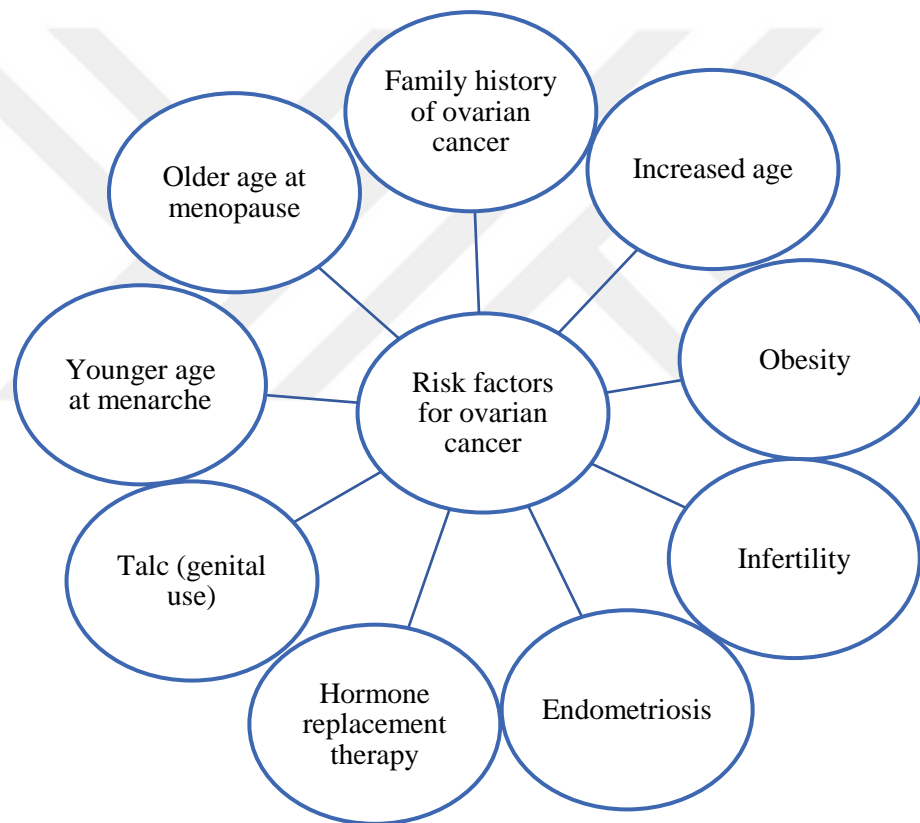


Figure 2 Risk factors associated with Ovarian Cancer

2.3. Epidemiology

Ovarian cancer incidence manifests large geographic variations (Figure 3) (WHO, 2018). The developed countries in North and Central America, and Eastern Europe demonstrated the highest (age-standardized) incidence rates, which rates are usually surpassing 8 per 100,000. Rates are generally moderate in Australia (> 6.7 per 100,000) and South America (~5.5 per 100,000) and

lowest in the Middle East with South Asia and Africa (≤ 3.8 per 100,000). OC rates were also increased in the less developed countries that undergo economic growth and lifestyle alterations. Immigration from less-developed to high-developed countries can also cause a shift in the rates of OC incidence and mortality, which indicates the influence of non-genetic risk factors (Herrinton, Stanford, Schwartz, & Weiss, 1994; Kliewer & Smith, 1995).

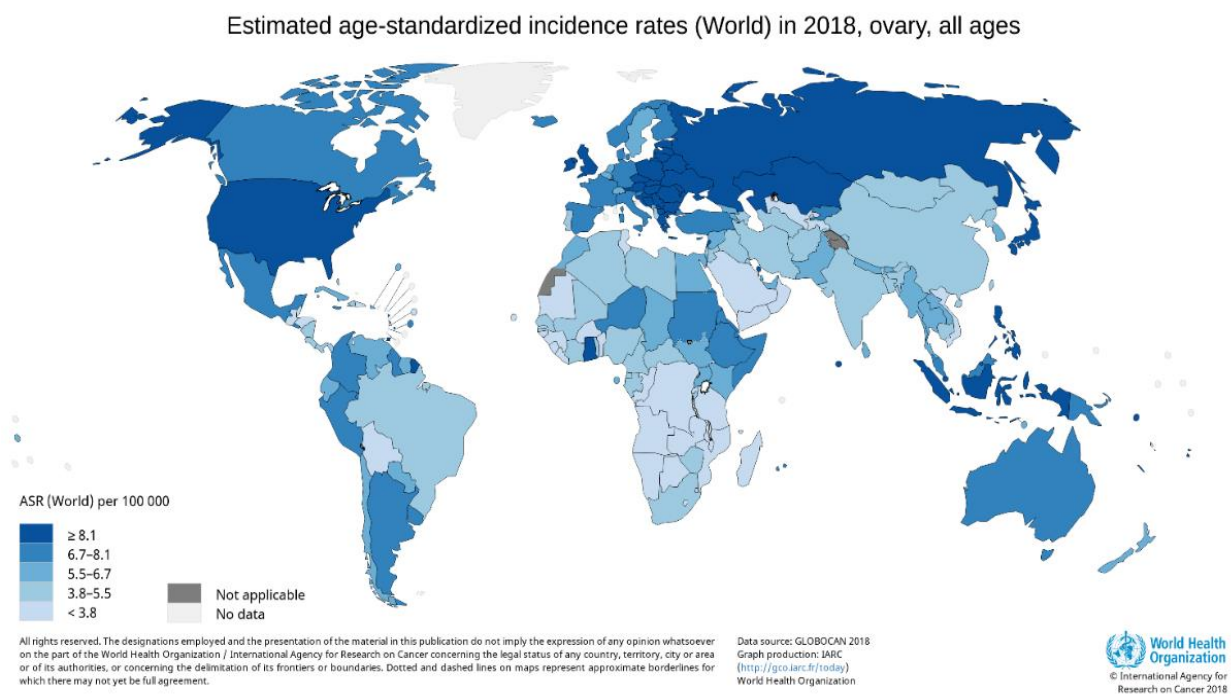


Figure 3 Global Incidence of Ovarian Cancer in 2018 (for all ages). ASR displays an age-standardized rate (WHO, 2018).

2.4. The Staging of Ovarian Cancers

The International Federation of Gynecology and Obstetrics (FIGO) ovarian cancer staging system was announced in 1973 for the first time and then adjusted in the years of 1988 and 2014 (Mutch & Prat, 2014). As it was showed in Table 1 and illustrated briefly in Figure 4, there are four stages to describe ovarian cancer based on the new FIGO classification system (Prat, 2015).

Table 1 FIGO Staging System for Ovarian Cancer (Adapted from Prat, 2015)

Stage IB	“Tumor on the surface of the either or both ovary or fallopian tube, capsule rupture, positive peritoneal ascites.”
Stage IIB	“Invasion to pelvic intraperitoneal tissues.”
Stage IIIA	“Metastasis to the lymph nodes or microscopic, peritoneal involvement with/out the involvement of lymph nodes.”
Stage IIIB	“Macroscopic, peritoneal metastasis equal or bigger than the size of 2 cm with/out the involvement of lymph nodes.”
Stage IIIC	“Macroscopic, peritoneal metastasis bigger than the size of 2 cm with/out the involvement of lymph nodes.”
Stage IV	“Metastasis to liver, spleen, or extra-abdominal organs.”

Stage I is relatively rare because patients are mostly diagnosed at stage III or IV, but it is associated with the most excellent survival rates (Richardson, Scully, Nikrui, & Nelson, 1985). Stage II represents the spread of the tumor into the different areas of the pelvis. It implies that it is not only in the ovaries, but it is also spreading into distinct tissues as well. This stage makes up less than 10 percent of tumors and is deemed curable. Unfortunately, owing to the lack of symptoms, the more significant part of women are generally diagnosed with OC at stage III or IV. Past studies reported that patients who showed non-specific symptoms such as abdominal pain, gastrointestinal symptoms, and bowel habits alterations might lead to a misdiagnose as if suffering from irritable bowel syndrome (Goff, Mandel, Munt, & Melancron, 2000; Olson et al., 2001). At late stages, the tumor has spread outside of the ovaries and pelvis. Stage IV is defined as a distant metastatic form of cancer, which accounts for twelve to twenty-one percent of ovarian cancer patients (Heintz et al., 2006).

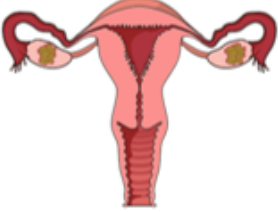



Stage I	Stage II
	
Tumor confined to one or both ovaries or fallopian tube(s)	Tumor continues to spread within the pelvic extension
Stage III	Stage IV
	
Tumor spreads to the peritoneum outside the pelvis	Distant metastasis occurs except peritoneal metastases

Figure 4 Stages of Ovarian Cancer (Figure created in the Mind the Graph platform, www.mindthegraph.com).

2.5. Cancer Treatment and Chemotherapy

There are three main types of cancer treatment: radiotherapy, surgery, and chemotherapy. Chemotherapy aims to use anti-tumor agents to cure cancer. The purpose is killing the cancer cells precisely without harming the healthy tissue.

The first recorded application of chemotherapy against cancer was to cure lymphoma by the nitrogen mustard in 1942 (Goodman et al., 1984). The significant landmarks of chemotherapy were the introduction of platinum in 1976 (Wiltshaw & Kroner, 1976), cisplatin-based therapy between 1984 and 1986 (Neijt et al., 1984; Williams et al., 1985; Omura et al., 1986), and paclitaxel in 1993 (Einzig et al., 1992; McGuire et al., 1996). These landmarks remarkably improved the survival in

females with advanced stages, and over the previous 20 to 30 years, their combined use has been refined through clinical trial involvement.

2.5.1. Platinum-based Chemotherapeutics

The achievement of platinum as an anti-tumor agent (Galanski, Jakupec, & Keppler, 2005) has inspired research for alternative metallic cytotoxic agents with advanced anti-tumor activity and less toxicity. Some of the antineoplastic agents based on platinum are currently in clinical use, such as cisplatin, carboplatin, and oxaliplatin (Figure 5) (Galanski, 2006). Cisplatin and carboplatin have been used to cure numerous cancers, such as testicular, ovarian, and neck (Muggia, 2009). Generally, these drugs bind at the N7 position of purines (Reedijk, 1987), which leads to unwinding and bending of the DNA (Johnstone, Suntharalingam, & Lippard, 2016). This action inhibits cellular division by preventing transcription and replication of DNA, which probably initiates apoptosis as a result of this action (Alderden, 2006; Amable, 2016).

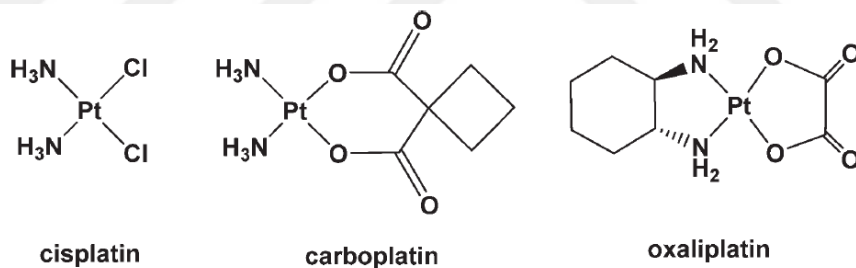


Figure 5 The structures of three platinum drugs that are currently in clinical use (Adapted from Wheate, Walker, Craig, & Oun, 2010).

Even though these heavy metal agents are effective against various cancer types (Muggia, 2009), they display serious side effects such as gastrointestinal symptoms, neuromuscular complications, and nephrotoxicity (Griffin et al., 1996). Furthermore, platinum use is limited by primary and chemoresistance in many tumor kinds (Brabec & Kasparikova, 2005; Chen, Milacic, Frezza, & Dou, 2009). These findings led to a continuous search for non-platinum metals that can expand the range of drug activity (Ott & Gust, 2007).

2.5.1.1. *Carboplatin*

Carboplatin exhibits a mechanism resembling cisplatin due to forming adducts with the DNA. Carboplatin also favors the N7 position of purines to bind. Nevertheless, carboplatin seems to be less reactive than cisplatin and exhibits less severe side effects. As cisplatin, it contains platinum enclosed by groups of two ammonia and two additional ligands in the cis position. The other two ligands in carboplatin have existed in a ring complex instead of two chlorides in cisplatin. This alteration makes carboplatin less reactive, stable, and lower side effects (nephrotoxicity, neurotoxicity, and ototoxicity) than cisplatin (Go & Adjei, 1999; McEvoy, 2000). A recent study proved that carboplatin favor to form monoadducts opposite to cisplatin, because of this, carboplatin requires a long time to form diadducts with the double-helix (Oliveira, Caquito, & Rocha, 2018).

Carboplatin has been used for many cancer types as ovary (Swenerton et al., 1992; Alberts et al., 1992), lung (Langer et al., 1995), and head tumors (Al-Sarraf et al., 1987), however, it has less efficiency compared to cisplatin in certain circumstances (Bajorin et al., 1993; Bokemeyer et al., 1996; Lokich & Anderson, 1998; Stewart, 2007).

2.5.2. *Non-Platinum-based Chemotherapeutics*

2.5.2.1. *Paclitaxel*

Paclitaxel is a chemotherapeutic emerges from the taxane class, which was extracted from the bark of the Pacific yew tree in the 1960s (Cragg, 1998; Suffness, 1994). As a microtubule agent, paclitaxel eliminates the microtubule spindle dynamics in the cell (Manfredi & Horwitz, 1984). The outcome of this is the inhibition of cell division and apoptosis induction. It is effective, particularly against breast, endometrial, bladder cancers, and cervical carcinoma (Hajek, Vorlicek, & Slavik, 1996). Paclitaxel treatment has historically given the first agent with constant activity in platinum-resistant patients (McGuire et al., 1989; Einzig et al., 1992; Trimble et al., 1993; Thigpen et al., 1994). However, the treatment with cisplatin and paclitaxel agents against advanced (stage IV) epithelial ovarian cancer was unsuccessful, and the combination of these drugs was not better than treatments of cisplatin or 24 h paclitaxel individually (David et al., 2007).

2.5.2.2. *Ruthenium*

Ruthenium complexes are another class of metal compounds. Since they also favor the same position of DNA (N7 of guanine) to bind, their mechanism of action might be similar to cisplatin (Clarke, 1989). Although the similarities between ruthenium and platinum, it is well known that ruthenium has many differences (Heffeter et al., 2008). First, ruthenium seems to favor neoplastic tissues to aggregate rather than healthy tissues, probably through transferrin usage to access to cancer cells (Sava et al., 1984). The study reported that Ru-transferrin complexes carried into the tumor masses. When the receptor (transferrin) attached, the complex detaches Ru, which is later taken into the cancer cell (Sava & Bergamo, 2000). Second, ruthenium maintains its comparatively passive Ru(III) oxidation condition until it arrives at cancer tissue, which has a lower pH and less amount of oxygen than healthy tissues. When it arrives at cancer tissue, reduction to a more active state Ru(II) takes place (Schluga et al., 2006). This reaction, named “activation by reduction”, provides a selective tumor targeting, which can be beneficial for hypoxic tumors that are possibly chemotherapy-resistant (Rockwell et al., 2009). Lastly, some of the Ru agents show better efficiency against cancer metastases (Gagliardi et al., 1994). Given these features, ruthenium is anticipated to demonstrate anti-cancer activity and cytotoxicity patterns that are different from those of platinum. These remarkable features may help to overcome platinum resistance together with an extended activity range.

Numerous structure-activity studies showed that three elements are critical to the cytotoxicity of Ruthenium compounds; chloride, N-N donor ligand, and arene ligand (Habtemariam et al., 2006; Ang & Dyson, 2006; Betanzos- Lara et al., 2012; Barry & Sadler, 2013). The Ru complexes that we used in this study containing these elements (Figure 10).

There are only two ruthenium agents, NAMI-A (New Anti-tumor Metastasis Inhibitor A) and KP1019, which introduced to the human clinical trials (Sava, Alessio, Bergamo, & Mestroni, 1999; Pieper, Borsky, & Keppler, 1999). NAMI-A is the first Ru agent that was used in human clinical trials. This Ru (III) complex is carrying an imidazole and DMSO (Dimethyl sulfoxide) coordinated to the ruthenium (Mestroni, Alessio, & Sava, 1998). Another Ru (III) complex, KP1019, contains ruthenium that is coordinated by two indazole heterocycles with nitrogen atoms (Figure 6).

Regardless of their similarities in the structural and chemical base, each agent demonstrates a different anti-tumor phenotype. In preclinical trials, NAMI-A has shown anti-metastatic activities in numerous animal models, but it seems it has no direct cytotoxic effects (Sava, Pacor, Mestroni, & Alessio, 1992; Sava et al., 1995). On the other hand, KP1019 has shown direct cytotoxic effects to an extensive range of primary tumors by promoting apoptosis (Galanski, Arion, Jakupec, & Keppler, 2003; Jakupec et al., 2005; Hartinger et al., 2008).

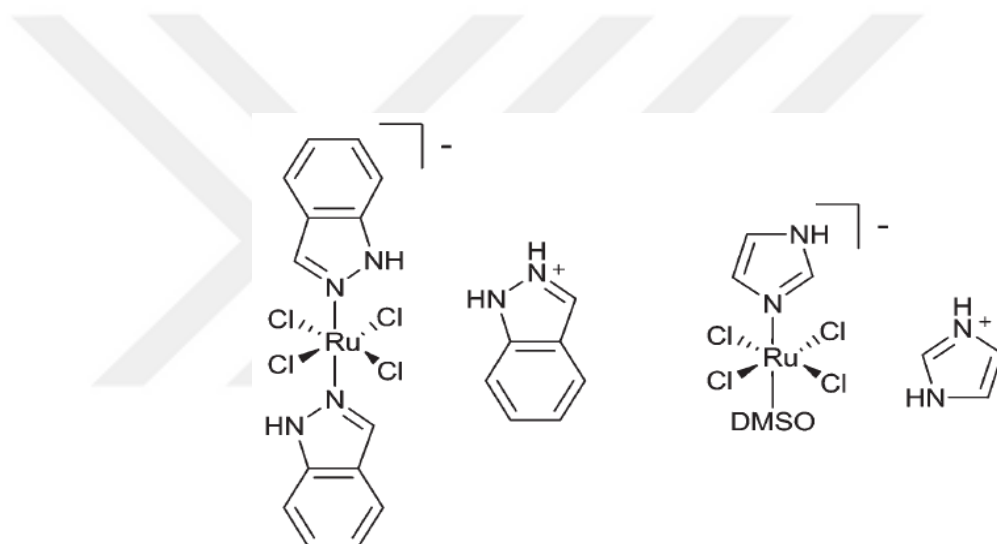


Figure 6 Structures of KP1019 and NAMI-A, respectively (Adapted from Alessio, Mestroni, Bergamo, & Sava, 2004).

A phase I study was started to use NAMI-A in 1999, and it was announced in 2004 (Rademaker-Lakhai et al., 2004). In the study, 24 patients bearing various metastatic tumors (such as colorectal, melanoma, and ovarian) were treated via dose increase protocol. 20 of 24 patients (83%) were successfully treated in this study. However, at high doses, they developed uncomfortable blisters on some body parts that persisted for weeks. Moreover, NAMI-A had other side effects such as anemia, lymphopenia, anorexia, stomatitis, and nausea. Interestingly KP1019 did not show any dose-associated side effects; however, its trial covered only eight patients, and the compound had complications with the solubility.

To confirm if the Ru agents were carried into the cells or not, the cellular uptake of complex 2, 1b, and 1a in OVCAR-3 cell line was examined in a study (Tavsan, Yaman, Subasi, & Ayar Kayalı, 2018). As can be seen in Figure 7, the overlay image demonstrated that the complex 1b efficiently absorbed by OVCAR-3 cells and able to enter into the cytoplasm. The aggregation was detected in cell nuclei. Complex 1a and 2 also showed the same manner (data not shown) as complex 1b. After 8 h, the cells were started to throw out the Ru compounds.

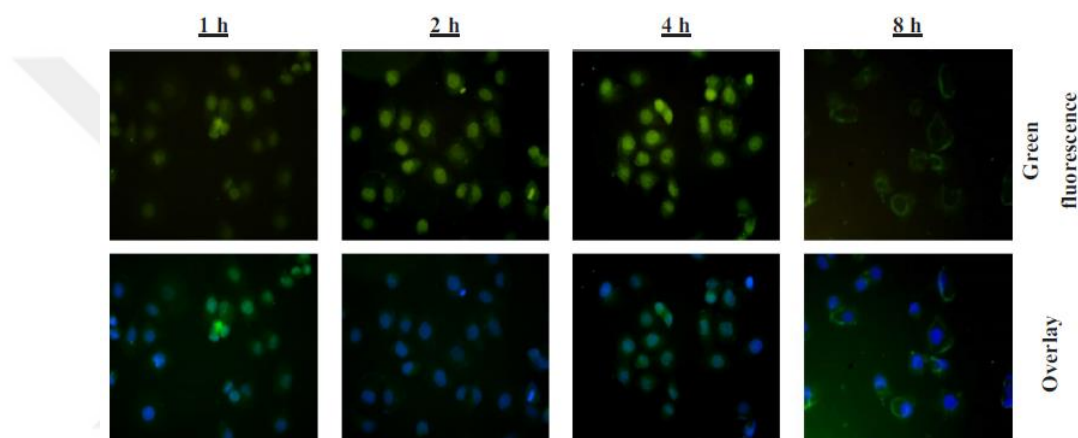


Figure 7 Images of 1b treated OVCAR-3 cells for variable incubation time. DAPI (blue) stain represents nuclei, the green fluorescence stain (green) represents complex 1b, and the overlay demonstrates the cellular relation of these compounds with DNA. (Adapted from Tavsan, Yaman, Subasi, & Ayar Kayalı, 2018).

In conclusion, Ru complexes are potential chemotherapeutics due to their extraordinary qualities allowing a selective entry into the cancer cells along with lower toxicity to healthy tissue. However, many features of these ruthenium agents still waiting to be elucidated for further use.

2.6. Metabolism in Cancer

Reprogrammed metabolism is recognized as a hallmark of cancer due to changed metabolic characteristics that have been seen in numerous cancers (Hanahan & Weinberg, 2011; Pavlova & Thompson, 2016). During the 1920s, Otto Warburg and his colleagues provided evidence for such adaptations when they determined increased glucose uptake by cancer cells (Warburg, Wind, & Negelein, 1927). Warburg et al. (1924) then reported that regardless of the presence of molecular oxygen, glycolysis to produce lactic acid was preferred by cancer cells as if they were simulating the conditions of anaerobic. (Warburg, Posener, & Negelein, 1924). It was called "aerobic

glycolysis" to precisely express this pathway in cancer cells (Warburg, 1956a; Warburg, 1956b). This event is also called "Warburg Effect". Even though oxidative phosphorylation (OXPHOS) provides a high amount of adenosine 5'-triphosphate (ATP), glycolysis supplies the essential biomolecule precursors that are needed by cancer cells to preserve their high proliferative state (Menendez & Lupu, 2007; Christofk et al., 2008).

Aerobic glycolysis does not cause the suppression of OXPHOS in cancer cells completely. Also, reducing glycolytic ATP production via inhibiting the pyruvate kinase's activity unsuccessful in stopping tumorigenesis, recommending that the primary function of glycolysis is not the ATP production (Israelsen et al., 2013).

The metabolic shift to aerobic glycolysis is an essential mechanism for cancer cells to synthesize proteins, nucleotides, and lipids. Glucose transporters (such as GLUT1) are overexpressed to compensate, and eventually, uptake of glucose is increased in tumor cells. Increased glucose uptake causes hyperglycemia, which leads to a more acidic and hypoxic environment. The relation between a low pH environment and gain of chemoresistance was examined in many tumor types (Kellenberger et al., 2010). Besides the regulation of glucose uptake and lactic acid production (Vander Heiden, Cantley, & Thompson, 2009; Wu & Zhao, 2013), there are other pathways in cellular metabolism that are reprogrammed in cancer cells such as increased glutamine metabolism and synthesis of fatty acid (Zhao, Butler, & Tan, 2013).

2.6.1. Glucose metabolism

The cells take glucose molecules inside through responsible transporters and then modify and phosphorylate it through a series of biochemical reactions (Berg, Combs, & Scherer, 2002). Glucose can proceed to lipid synthesis by transforming to glycerol or sequentially converted to pyruvate, which is called glycolysis. Glycolysis is a metabolic pathway that does not require molecular oxygen (O_2) and happens in the cytosol of both prokaryotic and eukaryotic organisms. By glycolysis, one glucose molecule breaks down into two pyruvate molecules with $NADH^+$ each, which yields two molecules of ATP. Another way to use glucose is nucleotide synthesis by "pentose phosphate pathway" or nicotinamide adenine dinucleotide phosphate (NADPH)

generation for reductive biosynthesis (Figure 8). Glucose is also necessary for glycosylation via “hexosamine biosynthetic pathway” (Wellen et al., 2010; Dang, 2012).

Oncogenic mutations result in the over-expression of GLUT (Macheda, Rogers, & Best, 2005; Pelicano, Martin, Xu, & Huang, 2006) hence assists the glucose consumption in tumor cells. Even though glycolysis generates a low amount of ATP than mitochondrial OXPHOS, there are certain advantages for tumor cells to prefer aerobic glycolysis (de Souza, Justo, de Araujo, & Cavagis, 2011). First, increased glycolysis rate and lactic acid production would lead to faster tumor growth. Pfeiffer et al. (2001) were proposed that fast but less ATP generation pathway offers particular benefits for circumstances of nutrient deprivation, adding an evolutionary magnitude to glycolysis (Pfeiffer, Schuster, & Bonhoeffer, 2001; Zhou et al., 2012). Second, cancer cells also require metabolic precursors that are necessary for the biosynthesis of molecules to support tumor growth (Deberardinis, Sayed, Ditsworth, & Thompson, 2008).

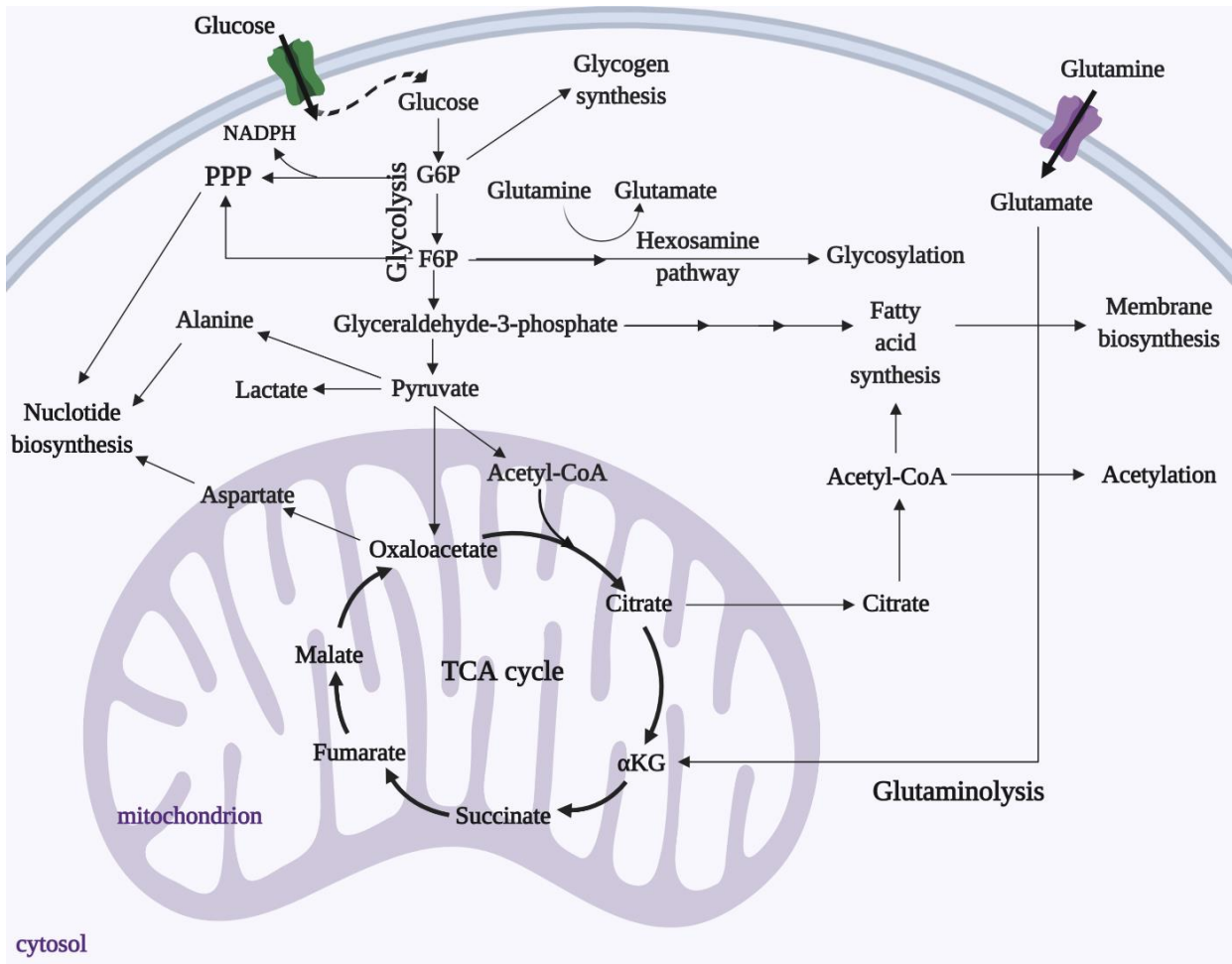


Figure 8 Energy metabolism in the cell (ATP and other cofactors have been removed for simplicity. G6P: glucose-6-phosphate, PPP: pentose phosphate pathway, F6P: fructose-6-phosphate, α KG: alpha-Ketoglutarate, TCA: tricarboxylic acid, NADPH: nicotinamide adenine dinucleotide phosphate. Figure created via biorender, www.biorender.com, and adapted from DeBerardinis & Chandel, 2016).

2.6.2. Tricarboxylic Acid Cycle

The TCA cycle is acknowledged as well as the Krebs or the citric acid cycle. Hans Krebs is the name behind the TCA cycle for his studies and also evident for the importance of glutamine metabolism in animals in 1935. The TCA cycle is an essential pathway in aerobic organisms, which is placed in the mitochondria. After glycolysis, the pyruvate molecule is transformed into acetyl-CoA. With the condensation of acetyl-CoA and OAA, the TCA cycle starts. Intermediates of the TCA cycle mainly consist of OAA, citrate, α -KG, succinyl-CoA, succinate, malate, and fumarate. These metabolites are used in numerous pathways, such as fatty acid synthesis, protein, and

nucleotide biosynthesis (Ahn & Metallo, 2015). TCA cycle produces electrons, carbon skeletons, and CO₂ that can be used to replenish the cycle, which is called “anaplerosis”. Citrate can be transformed into acetyl-CoA in the cytosol for the membrane biosynthesis. Oxaloacetate can be transformed to aspartate to use for nucleotide synthesis (Son et al., 2013). Moreover, αKG can be transformed into citrate by reversing the TCA cycle by “reductive carboxylation”, that later use in the creation of acetyl-CoA and then fatty acid synthesis to support membrane biosynthesis (Ward et al., 2010). These pathways mainly illustrated in Figure 8.

Mitochondrial dysfunction is commonly observed in cancer cells and crucial for tumor progression (Hu et al., 2012; Lu et al., 2012). Gene mutations in mitochondrial DNA (mtDNA) alter the bioenergetics of cancer cells and help to adapt to environmental changes (Brandon, Baldi, & Wallace, 2006). Succinate dehydrogenase (SDH) and fumarate hydratase (FH) mutations are frequently seen in various types of cancers (Bayley & Devilee, 2010). Therefore it is clear that mitochondrial dysfunction may cause a shift in the cellular metabolism of cancer cells.

2.6.3. *Electron Transport Chain*

The electron transport chain (ETC) consists of four respiratory complexes that are responsible for the transportation of electrons by redox reactions and also pumps the protons (H⁺ ions) into the inner mitochondrial membrane. Molecular oxygen (O₂) has a vital role as the last acceptor of electrons in aerobic respiration. Electron flow creates an electrochemical proton gradient that utilizes a driving force for ATP synthase to phosphorylate ADP (Mitchell, 1961; Nicholls & Budd, 2000; Wallace, 2013). Several metabolic pathways (such as glycolysis and TCA cycle) supply electron donors to ETC.

ETC is identified as an essential pathway that controls signaling, biosynthetic pathways, and bioenergetics during proliferation and metastasis of cancer cells. A study suggests that the crucial role of the ETC in cell proliferation is to support the conversion of aspartate (Birsoy et al., 2015). Aspartate can be used for nucleotide synthesis to support cell proliferation (Son et al., 2013).

Even though the severe inhibition of ETC showed anti-growth properties, the temperate inhibition of ETC that contains heteroplasmic mitochondrial DNA mutations can benefit to

tumorigenesis, possibly by raising the reactive oxygen species levels (Kulmacz, 1989; Petros et al., 2005; Ishikawa et al., 2008).

2.6.4. Anaerobic Respiration

Anaerobic respiration uses electron acceptors that are different from O₂. Even though oxygen has not been using, the reaction process in a respiratory electron transport chain (Slonczewski & Foster, 2011). Other electron acceptors in anaerobes possess lower reduction potentials, thus result in less energy release for each oxidized molecule.

Although highly proliferative cancer cells display high glucose uptake, most of the glucose is transformed into lactate rather than consumed in OXPHOS (Mayers & Vander Heiden, 2015). Raised lactate production leads to less pyruvate availability for the TCA cycle. However, with this conversion, the cells can reproduce NAD, which is needed in the glycolysis pathway.

2.6.5. Amino Acid Metabolism

The non-essential amino acid glutamine as an important anaplerotic substrate can be used in the conversion of glutamate to α KG. Thus it can replenish the TCA cycle via glutaminolysis (Hensley, Wasti, & DeBerardinis, 2013). The cancer cells that are supplied with α KG, oxaloacetate, or pyruvate is sufficient for them to survive in glutamine deprivation conditions, confirming that glutamine assists tumor growth by replenishing TCA cycle intermediates via anaplerosis (Yuneva et al., 2007; Weinberg et al., 2010; Altman, Stine, & Dang., 2016). As reverse reaction can happen, the transformation of alanine to pyruvate and aspartate to oxaloacetate can happen to support the TCA cycle or glycolysis when it is necessary.

Glutamine and glutamate have a fundamental part in non-essential amino acid (NEAA) metabolism, and each can synthesize the other NEAAs (Choi & Coloff, 2019). Many cancer cell types showed a glutamine dependent behavior to sustain their energy needs (Wise & Thompson, 2010). The conversion of glutamine to glutamate happens via the glutaminase enzyme. Glutamate supplies the TCA cycle and assists in the biosynthesis of fatty acids (Wise et al., 2008; Mullen et al., 2012). The studies showed that the tumor-initiating cells were able to endure the glucose deficiency on glutamine-supplemented media due to glutamine usage in the TCA cycle via

glutaminolysis (Roberts et al., 2005; Anderson et al., 2014). It has been proved that hypoxia does not weaken the glutamine metabolism in proliferating cells. Glutamine can supply to fatty acid biosynthesis by reversing the TCA cycle or reductive carboxylation of α -ketoglutarate to citrate via isocitrate dehydrogenase (IDH) (Wise et al., 2011; Metallo et al., 2012; Mullen et al., 2012).

Hudson et al. (2016) observed that the overexpression of glutaminase and more sensitivity to glutamine deprivation in platinum-resistant ovarian cancer cells compared to platinum-sensitive ovarian cancer cells. It was suggested that increased glutamine consumption is an important step to acquire platinum resistance (Hudson et al., 2016).

2.6.6. *Metabolism in Ovarian Cancer*

The studies proved that the Warburg effect is highly essential for the development of ovarian cancer cells (Deberardinis, Sayed, Ditsworth, & Thompson, 2008; Caneba et al., 2014). Another study proposed that the primary purpose of the Warburg effect is to keep glycolytic intermediates at a high amount to supply for the biosynthesis of macromolecules, which is preferred by fast-cycling cancer cells (Vander Heiden, Cantley, & Thompson, 2009). On the contrary, slow-cycling cells mostly favor OXPHOS as a preferred energy source (Roesch et al., 2013).

In ovarian cancer cells, glucose is broadly used to produce ATP and sustain energy and redox balance (Vander Heiden et al., 2010; Ahn & Metallo, 2015). The study of Ippolito et al. (2016) has found that chemoresistant cells are excellent miners of glucose (Ippolito et al., 2016). Additionally, another study proved that cisplatin-resistant ovarian cancer cells showed higher glucose demand, and they were more delicate to glucose deficiency (Catanzaro et al., 2015). However, Xu et al. (2018) proved that cisplatin-resistant ovarian cancer cells had a greater demand for glucose and demonstrated higher glucose uptake and consumption with decreased lactate production (Xu et al., 2018). Moreover, they were less delicate to glucose deficiency due to their greater stocks of glycogen. These cells displayed exceptional reduces in extracellular lactate and ECAR (extracellular acidification rate - an indicative of glycolysis). Dar et al. (2017) also found that chemoresistant ovarian cancer cells could endure the conditions of limiting glucose and they demonstrated higher OXPHOS activity (Dar et al., 2017). These findings are recommending that the energy metabolism of ovarian cancer cells may differ from one cell line to another.

Accumulation of cellular glycogen was recognized as an aspect of clear cell ovarian carcinoma, which usually generates chemoresistance (Iida et al., 2012). Additionally, glycogen accumulation was increased under hypoxic circumstances (Uekuri et al., 2013). These findings indicate that particular metabolic phenotypes advance the chemoresistance in ovarian cancer.

Fatty acid metabolism was also changed in tumors. In ovarian cancer, unsaturated lipids were elevated (Zhang, Zhang, & Cheng, 2015), which stimulates stemness, while lipid desaturation impairs tumor initiation and cancer progression (Li *et al.*, 2017).



3. MATERIALS AND METHODS

3.1. Type of Research

The type of research performed in this study is experimental.

3.2. Time and Place of Research

The experiments performed at Ayar Kayalı Lab at Izmir Biomedicine and Genome Institute between October 2018 and May 2019.

3.3. Study Plan and Calendar

The experimental outline of this study summarized in Figure 9.

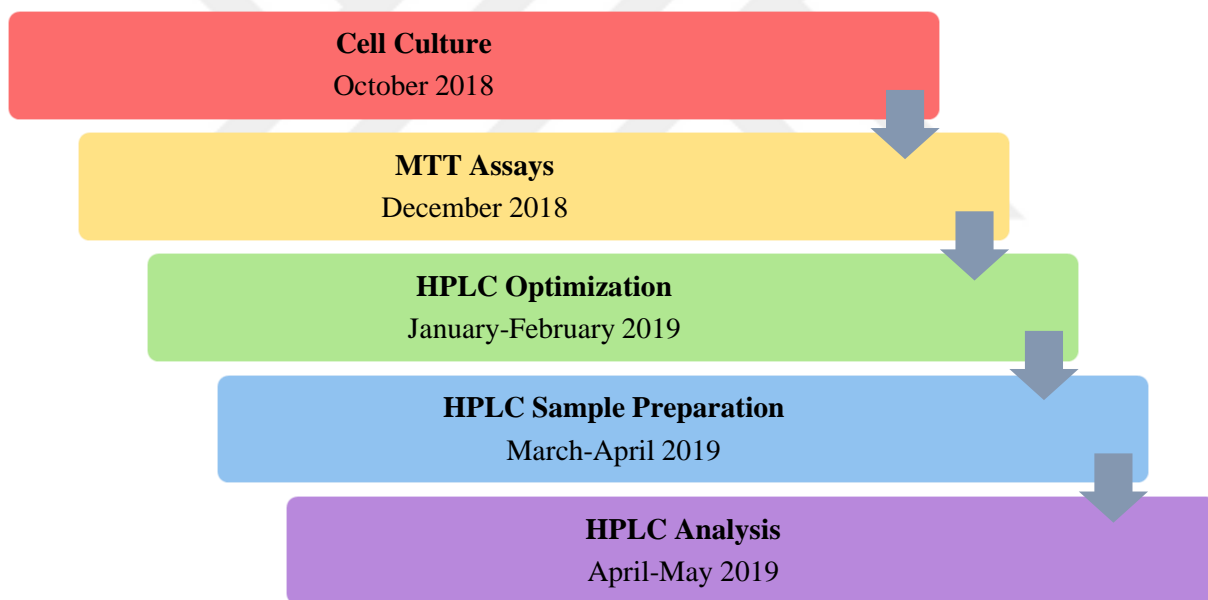


Figure 9 Outline of the study.

3.4. Materials

DNPH was obtained from Sigma-Aldrich (Sternheim, Germany). All standards were High-Performance Liquid Chromatography (HPLC) grade. The synthesis of TSC¹, TSC², and ruthenium complexes, $[(\eta^6\text{-}p\text{-cymene})\text{Ru}(\eta^1\text{-}S\text{-TSC})\text{Cl}_2]$ (1b), $[(\eta^6\text{-}p\text{-cym})\text{Ru}(\eta^1\text{-}S\text{-TSC}^2)\text{Cl}_2]$ (2) (Yaman, Şen, Karagöz, & Subaşı, 2017), and trans- $[\text{RuCl}_2(\text{PPh}_3)_2(\eta^2\text{-}N,S\text{-TSC})]$ (1a), were described before (Tavsan, Yaman, Subasi, & Ayar Kayalı, 2018). All ruthenium agents (Figure 10) were

kindly supplied by E. Subasi, Department of Chemistry, Faculty of Science, Dokuz Eylül University. Carboplatin (Carbodex®, 150 mg/15 ml) and paclitaxel (Ataxil®, 100 mg/16.7 ml) (Kocak Farma, Istanbul, Turkey) were obtained from a local pharmacy. The chemicals used in this study, as in the following Table 2.

Table 2 The chemicals used in this study.

Chemicals	Vendors
DNPH	Sigma-Aldrich
Oxaloacetic acid	Sigma-Aldrich
Lactic acid	Sigma-Aldrich
Succinic acid	Sigma-Aldrich
α -Ketoglutaric acid	Sigma-Aldrich
Fumaric acid	Sigma-Aldrich
Pyruvic acid	Sigma-Aldrich
Sodium Hydroxide (NaOH)	Sigma-Aldrich
Hydrochloric acid (HCl, 37%)	Sigma-Aldrich
H ₂ SO ₄	Sigma-Aldrich
Isopropanol (IPA)	Sigma-Aldrich
Malic acid	Merck
D-(+)-Glucose monohydrate	Merck
Ethanol	Isolab
Carbodex® and Ataxil®	Koçak Farma

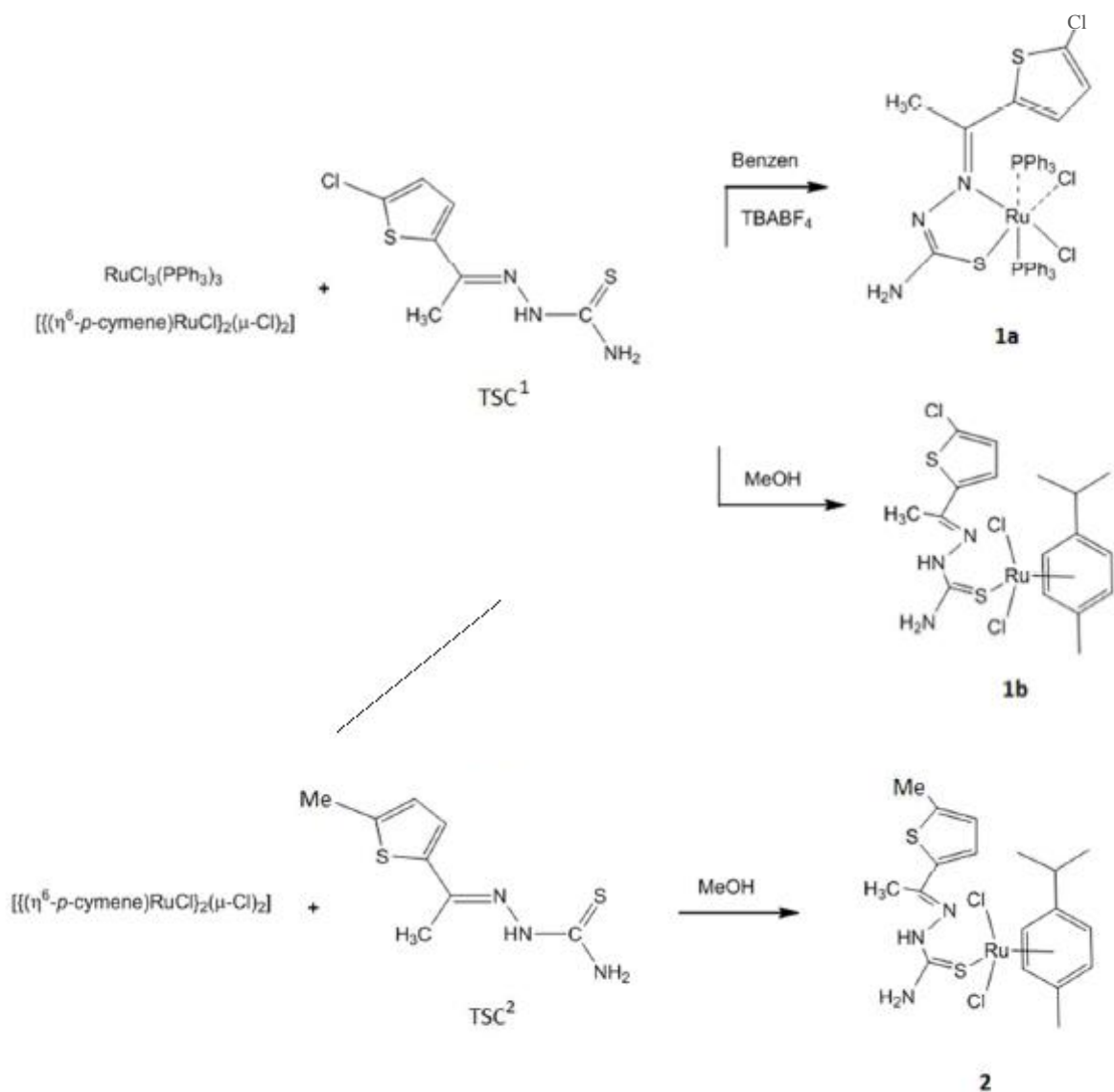


Figure 10 Ru (III) complexes. Complex 1a, 1b, and their ligand are TSC¹, Complex 2, and its ligand is TSC² (Adapted from Tavsan, Yaman, Subasi, & Ayar Kayalı, 2018).

3.4.1. Machines

The machines used in this study as in the following Table 3.

Table 3 The machines used in this study.

Machines	Vendors
Centrifuge 5810R	Eppendorf

Multiskan™ GO Microplate Spectrophotometer	Thermo Scientific
Centrifuge MicroCL 17R	Thermo Scientific
Balance - basic	Sartorius
HPLC System	Shimadzu
StabiliTherm™ Ovens and Incubators	Thermo Scientific
Axio Vert.A1 Inverted Microscope	ZEISS
Vortex Mixer	Thermo Scientific
Water Bath and Lid	Nüve
Dry Bath Incubator	Allsheng
pH Meter	Hanna

3.4.2. Cell Culture Materials

The cell lines were OSE (human normal ovarian surface epithelium), OVCAR-3, A2780, and A2780cis. Immortalized OSE cell line by SV-40 transfections was purchased from Abm-Good. A2780, A2780cis, and OVCAR-3 cell lines were purchased from the ECACC. Cell culture plates, flasks, cryovials, centrifuge tubes, filter tips, and serological pipettes were purchased from Isolab (Germany). Used materials in cell culture, as in the following Table 4.

Table 4 Used Materials in Cell Culture.

Materials	Vendors
DMEM	Sigma-Aldrich
DMSO	Sigma-Aldrich
PBS	Gibco
Trypsin-EDTA	Gibco
FBS	Gibco
Penicillin (100 units/ml)- Streptomycin (100 mg/ml)	Gibco

RPMI-1640	Sigma-Aldrich
MTT	Sigma-Aldrich

3.5. Methods

3.5.1. Cell Culture Conditions

A2780 (ovarian cancer cell line sensitive to cisplatin), A2780cis (ovarian cancer cell line resistant to cisplatin), and OVCAR-3 (ovarian cancer cell line resistant to cisplatin) were grown in RPMI-1640 medium. OSE-SV40 cell line was grown in DMEM growth medium. All growth medium supplemented with 10% fetal bovine serum (FBS) and 1% penicillin-streptomycin. The cells were kept in the incubator provides an atmosphere of 5% CO₂ at 37 °C.

3.5.2. Cell Viability Assay

3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) assay (Mosmann, 1983) was used to assess the mitochondrial dehydrogenase activity in the cells. The cells (100 µL/well) were seeded at a density of 7500 cells per well into 96-well plates and allowed them to grow for 24 h. In all assays, the concentration range of complex 1a, 1b, 2, and carboplatin were between 0-100 µM, and the concentration range of paclitaxel was between 0-100 nM. All Ru agents were dissolved in DMSO then diluted in PBS solutions serially. Later, they were diluted in the culture medium. All cells were treated for 24 h and 48 h. During the inclusion of the ruthenium complexes onto the cells, compounds were protected from light. Finally, 100 µL of MTT (5 mg/mL) was added on the cells and then incubated for 3 to 4 h in the incubator. The formazan crystals were dissolved via DMSO, and the absorbance was measured at 540 nm in Multiskan™ GO Microplate Spectrophotometer. The IC₅₀ values (concentration of agent causing 50% inhibition of growth) were calculated by curve fitting. IC₅₀ values of compounds were used later to treat the cells (for 24 h and 48 h) before collecting samples for HPLC analysis.

3.5.3. Preparation of Samples for HPLC Analysis

A modified protocol (Ganzera et al., 2006) was used for sample preparation. Each cell line was grown to 70% confluency before they were collected. The metabolites were extracted by adding boiling ultra-pure water on frozen cell pellets. The mixture was put inside in a boiling water

bath for 20 min and shook every 5 min. After cooling the sample, it was centrifuged for 20 min at 13,000 rpm. The supernatant was centrifuged once more before HPLC analysis. Extracellular metabolites (such as extracellular glucose) were extracted by collecting the medium of the cells after treatment of 24 h and 48 h. Then, the medium was centrifuged the same as the previous method for intracellular metabolites.

3.5.4. HPLC Conditions for Metabolites

The analysis of organic acids was examined by Shimadzu prominence HPLC system (Kyoto, Japan) equipped with an Alltech IOA-1000 column and a UV detector and a refractive index detector (RID). The conditions were the following: mobile phase 9.0 mM H₂SO₄ solution; injection volume, 20 μ L; flow rate, 0.4 mL/min; detection wavelength, 210 nm, and column temperature, 42 °C. Standards were prepared for OAA, α -KG, succinate, fumarate for UV detector, and for glucose, lactate, malate for RID, and the calibration curves were generated (Ayar Kayalı, 2005).

3.5.5. Colorimetric Determination of Pyruvic Acid

The concentration of pyruvate was assessed via 2, 4-dinitrophenylhydrazine (Aras & Ersen, 1975). The protocol was modified and recalculated to apply for smaller, 100 μ L, amount (96 well plate). 50 mg 2,4-dinitrophenylhydrazine (DNPH) was dissolved in 50 mL 2N Hydrochloric Acid (HCl, 37%). 30 μ L sample added into wells, and then 10 μ L DNPH was added to each solution and gently shook for 5 min at room temperature. 60 μ L 2N Sodium Hydroxide (NaOH) was added on the mixture and again shook for 10 min at room temperature. Absorbance was measured at 520 nm. Blank and standard solutions of pyruvic acid were prepared in the same way. For the calibration curve, 1.5 mg pyruvic acid was dissolved in 3 mL distilled water (0.5 mg/mL=50x10⁻² mg/mL). Then it was diluted to concentrations of 0.25x10⁻² mg/mL, 0.5x10⁻² mg/mL, 0.10x10⁻² mg/mL, 2.5x10⁻² mg/mL, 5x10⁻² mg/mL, 10x10⁻² mg/mL solutions. The calibration curve was created for pyruvate by plotting the concentration against the corresponding absorbance, which was used for the detection of the pyruvate amount in samples.

3.5.6. *Statistical analysis*

Each experiment was conducted at least three times with close outcomes and performed separately in triplicates. The results are average \pm S.D of at least three independent experiments.



4. RESULTS

4.1. Cytotoxic Activities of Chemotherapeutic Agents on Ovarian Cell Lines

The aim of the MTT assay was to assess the cytotoxic activity of each agent against human ovarian cell lines (OSE, A2780cis, A2780, and OVCAR-3). Cytotoxic activities of ligands, carboplatin, and paclitaxel were also tested under the same conditions to serve as a comparison. MTT assay was carried out by testing the agents in the concentration range of 0-100 μM , but the concentration range of paclitaxel was 0-100 nM due to its high cytotoxicity. All MTT assay studies conducted for 24 h and 48 h. The inhibitory potency (IC_{50}) (the concentration of agent reduced the cell viability by half) used to express the cytotoxicity.

Figure 11 displays the results of complex 1a treatment in ovarian cell lines for 24 h and 48 h. Inhibition in OSE cells reached the maximum value (90%) at 25 μM at 48 h, and 24 h line reached 88% inhibition at 25 μM (Figure 11A). A2780 cells showed a more delayed response to complex 1a compared to OSE and 89% inhibition observed at 20 μM at 48 h even though 24 h values failed to show the same; it was 70% at 20 μM (Figure 11B). A2780cis cells demonstrated more resistance than A2780 cells, but it was successfully reached to inhibition of 79% at 10 μM at 48 h, and for 24 h line, it was 55% inhibition (Figure 11C). OVCAR-3 cells showed a more resistant profile to treatment compare to other cell lines and reached 68% inhibition at 50 μM at 24 h. It showed 80% inhibition at 48 h at the same dose (Figure 11D). In general, results of 24 h applications were overrun by 48 h in all cell lines (OSE, A2780, A2780cis, OVCAR-3) because of increased cell death at 48 h and high toxicity of complex 1a (Figure 11).

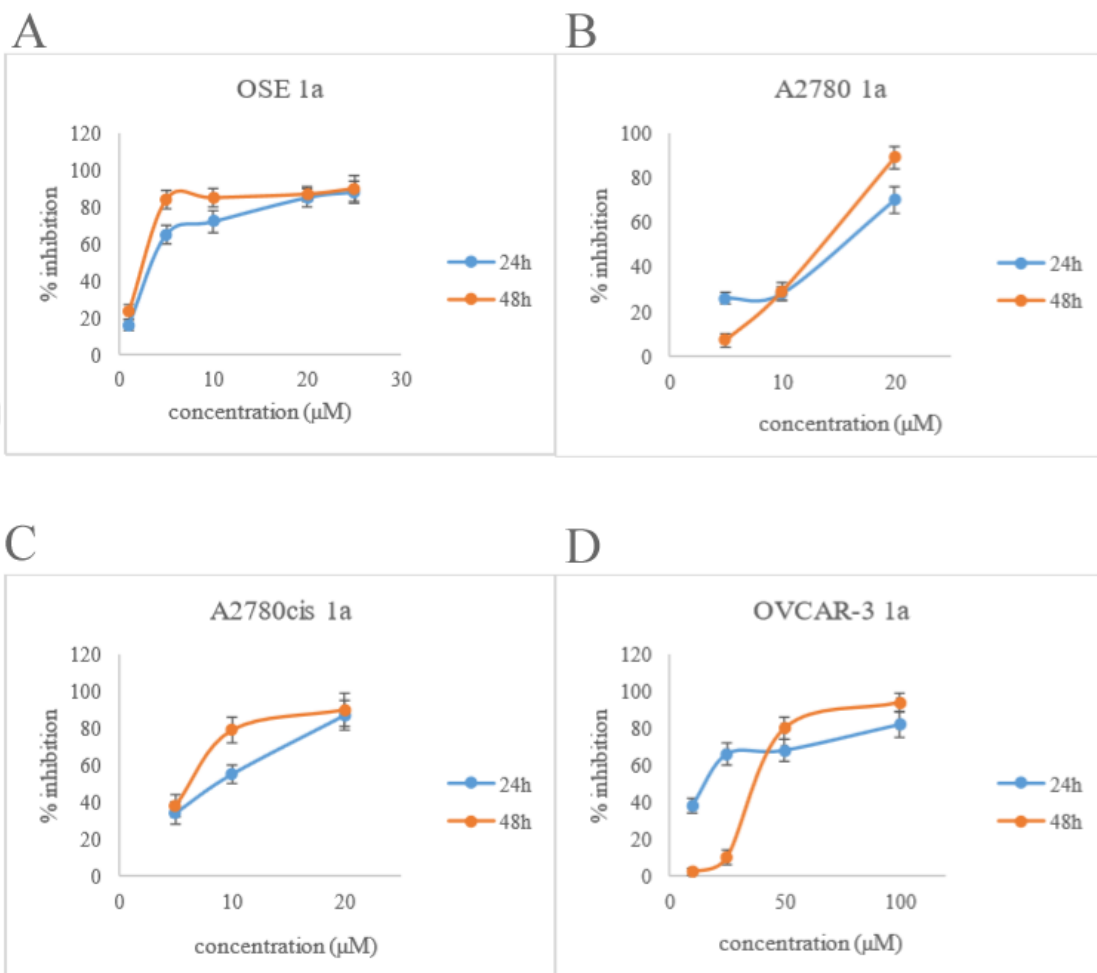


Figure 11 MTT results of complex 1a treatment in ovarian cell lines (A: OSE cell line with treatment of complex 1a for 24 h and 48 h; B: A2780 cell line with treatment of complex 1a for 24 h and 48 h; C: A2780cis cell line with treatment of complex 1a for 24 h and 48 h; D: OVCAR-3 cell line with treatment of complex 1a for 24 h and 48 h).

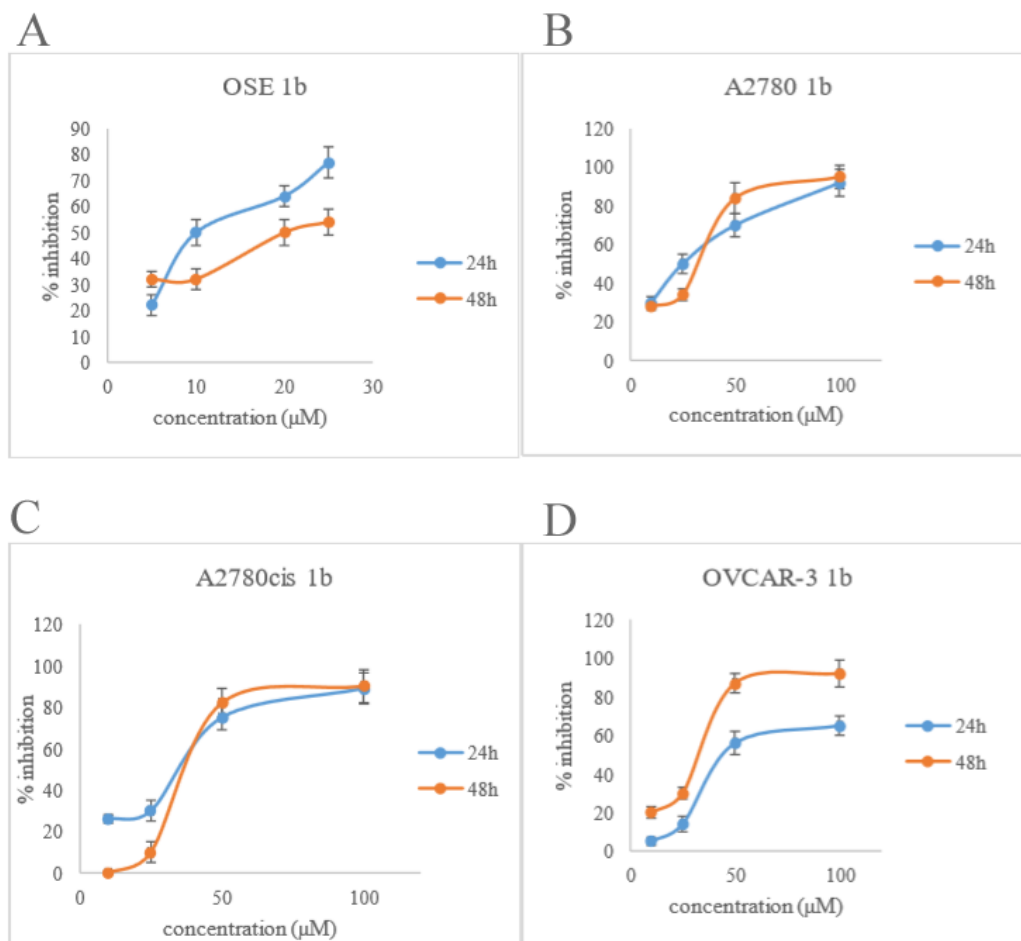


Figure 12 MTT results of complex 1b treatment in ovarian cell lines (A: OSE cell line with treatment of complex 1b for 24 h and 48 h; B: A2780 cell line with treatment of complex 1b for 24 h and 48 h; C: A2780cis cell line with treatment of 1b complex for 24 h and 48 h; D: OVCAR-3 cell line with treatment of complex 1b for 24 h and 48 h).

OSE cells that include complex 1b applied medium passed 50% inhibition after 10 µM at 24 h. Nevertheless, it was reached 50% inhibition at 20 µM at 48 h. Also, it was showed an increased cell number due to decreased drug toxicity at 48h (Figure 12A). Complex 1b applied A2780 cells displayed 70% inhibition at 50 µM at 24 h and 84% inhibition at 48 h (Figure 12B). A2780cis reached to the maximum inhibition value (89-90%) at 100 µM at 24 h and 48 h (Figure 12C). Complex 1b treatment in OVCAR-3 cells was highly effective, even at 48 h. It was reached to inhibition of 65% and 92% at 100 µM at 24 h and 48 h, respectively (Figure 12D).

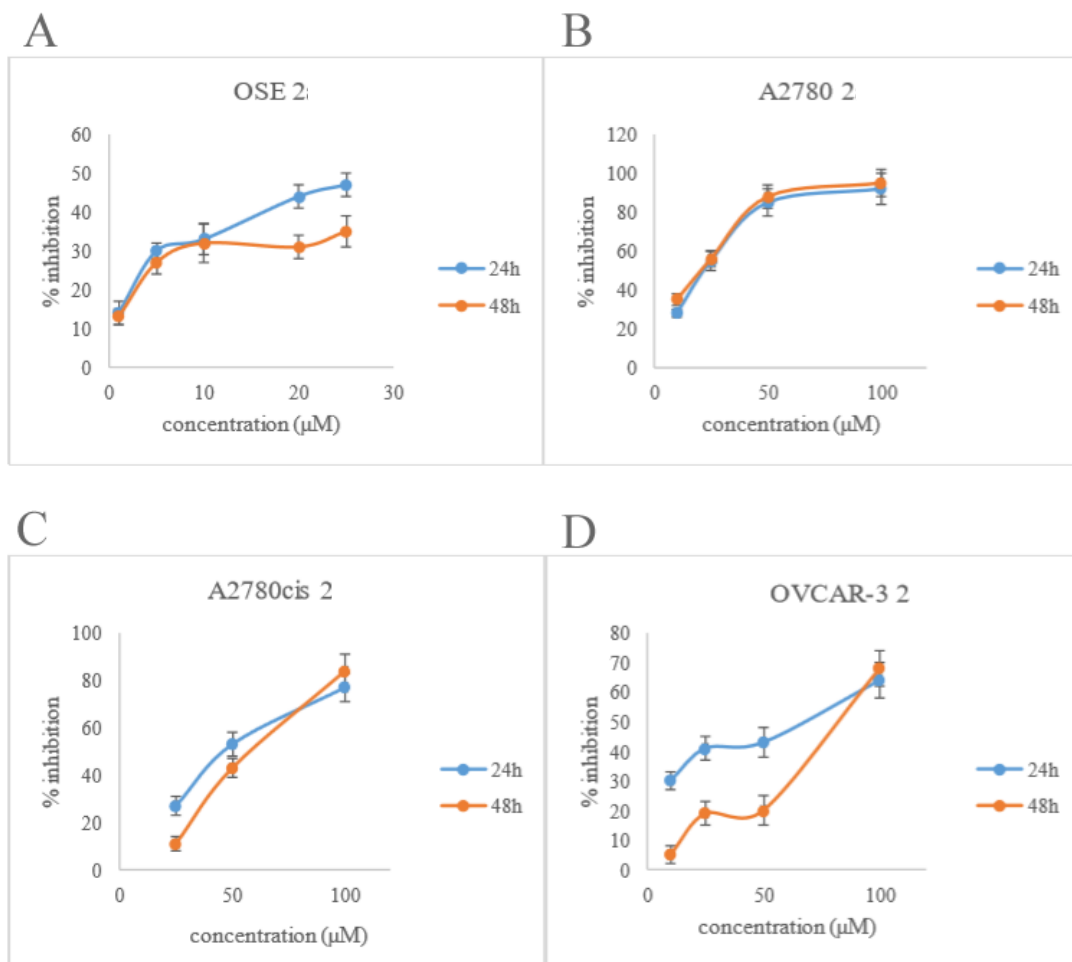


Figure 13 MTT results of complex 2 treatment in ovarian cell lines (A: OSE cell line with treatment of complex 2 for 24 h and 48 h; B: A2780 cell line with treatment of complex 2 for 24 h and 48 h; C: A2780cis cell line with treatment of complex 2 for 24 h and 48 h; D: OVCAR-3 cell line with treatment of complex 2 for 24 h and 48 h).

Complex 2 applied OSE cells failed to reach 50% inhibition between 1-25 µM doses, and toxicity was decreased at 48 h. The highest inhibition value (47%) was obtained at 25 µM at 24 h and 35% at 48 h for the same dose (Figure 13A). Nevertheless, A2780 cells reached a maximum inhibition value (92%) at 100 µM at 24 h and 94% at 48 h (Figure 13B). A2780cis cells passed 53% inhibition after 50 µM at 24 h and reached 84% inhibition at 100 µM at 48 h (Figure 13C). Complex 2 applied OVCAR-3 cells were able to reach 64% inhibition at 100 µM at 24 h and 68% at 48 h. However, complex 2 efficiency was generally lower at 48 h compared to 24 h in OVCAR-3 cell line (Figure 13D).

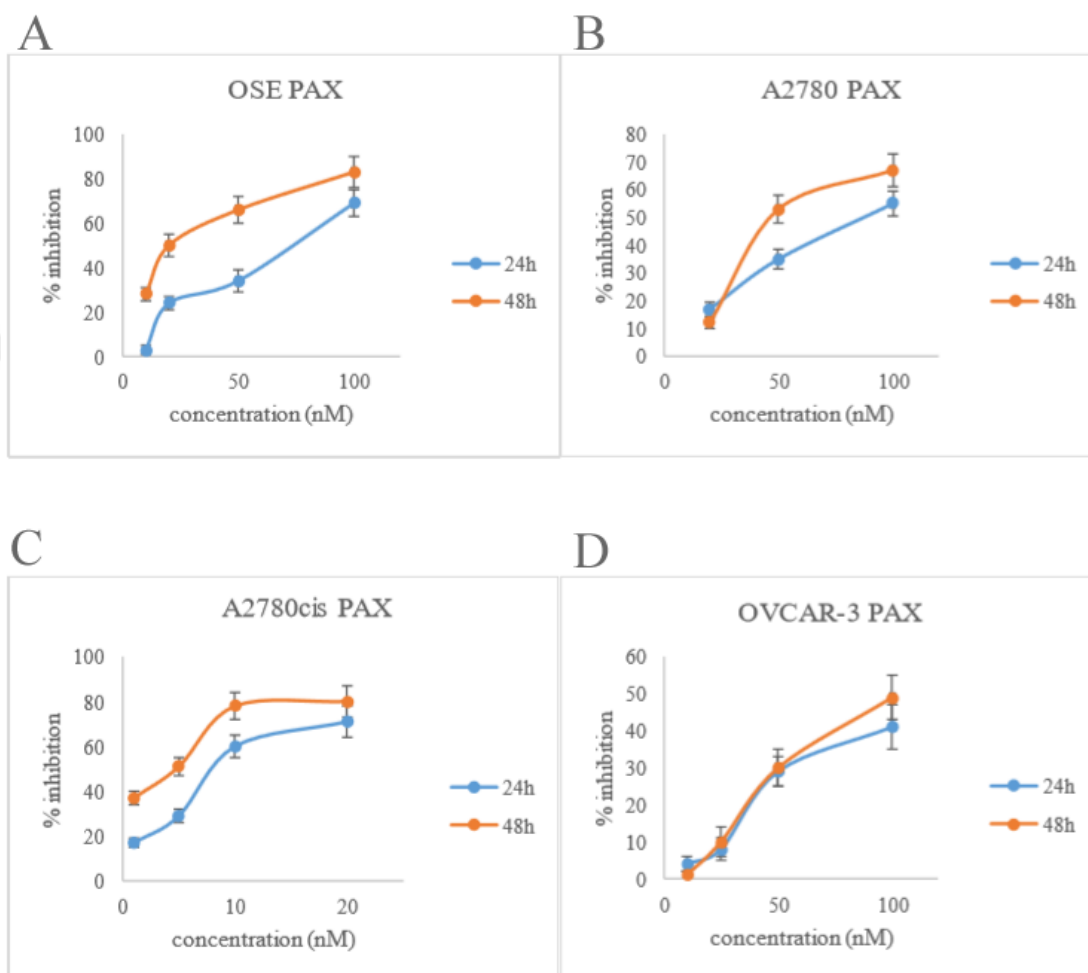


Figure 14 MTT results of PAX treatment in ovarian cell lines (A: OSE cell line with treatment of PAX for 24 h and 48 h; B: A2780 cell line with treatment of PAX for 24 h and 48 h; C: A2780cis cell line with treatment of PAX for 24 h and 48 h; D: OVCAR-3 cell line with treatment of PAX for 24 h and 48 h).

OSE cells showed a linear-like line after PAX treatment, and cytotoxicity was increased greatly at 48 h. It obtained 69% inhibition at 100 nM at 24 h and 83% inhibition at 48 h for the same dose (Figure 14A). A2780 cells with PAX treatment displayed 55% inhibition at 100 nM at 24 h and 53% inhibition at 50 nM at 48 h (Figure 14B). Inhibition value of PAX applied A2780cis cells was 71% inhibition at 20 nM at 24 h and 78% inhibition at 10 nM at 48 h (Figure 14C). On the other hand, OVCAR-3 showed 41% inhibition at 24 h and 49% at 48 h. (Figure 14D). In general, PAX displayed increased cytotoxicity to all cell lines at 48 h compared to 24 h.

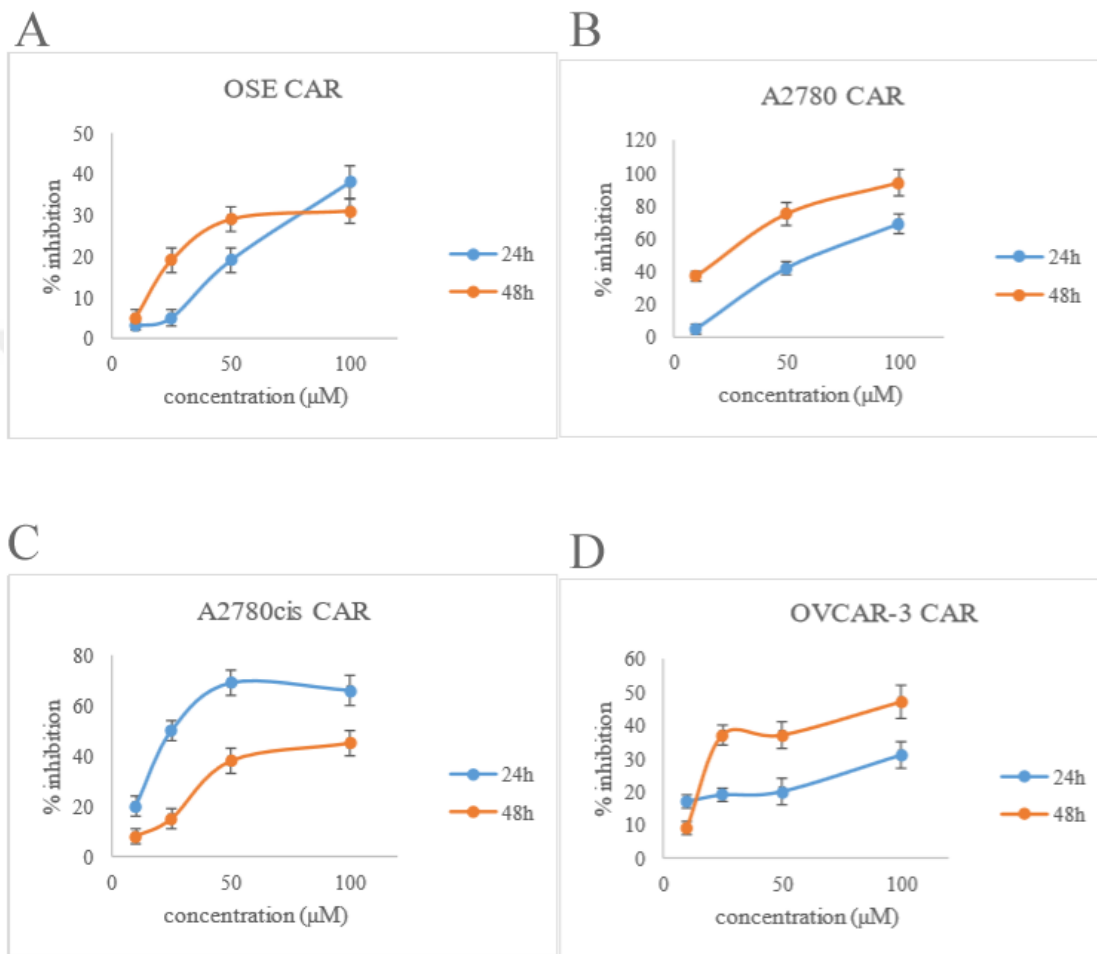


Figure 15 MTT results of CAR treatment in ovarian cell lines (A: OSE cell line with treatment of CAR for 24 h and 48 h; B: A2780 cell line with treatment of CAR for 24 h and 48 h; C: A2780cis cell line with treatment of CAR for 24 h and 48 h; D: OVCAR-3 cell line with treatment of CAR for 24 h and 48 h).

CAR applied OSE cells showed an increasing cytotoxicity profile and obtained 38% inhibition at 100 μM at 24 h while it was reached to maximum inhibition value (31%) at 48 h (Figure 15A). Although CAR treatment in A2780 cells showed only 69% inhibition at 100 μM at 24 h, it reached 94% inhibition at 48 h (Figure 15B). CAR application in A2780cis cells was more effective at 24 h than 48 h. The inhibition value was maximum (66%) at 100 μM at 24 h and 45% at 48 h (Figure 15C). As can be seen in Figure 15D, OVCAR-3 cells were more resistant to CAR, thus failed to reach to 50% inhibition value between 10-100 μM, although it was more effective at 48 h than 24 h. It was obtained %31 and %47 inhibition at 100 μM at 24 h and 48 h, respectively.

Table 5 demonstrates the IC₅₀ values of compounds that we used in MTT assays for 24 h. The observed range of IC₅₀ values for all compounds was between 11.6 ± 1.7 nM (PAX) and 234.5 ± 7.7 μM (CAR) towards tumor cell lines. Between all agents, PAX was the most toxic one and not only to cancer cells (A2780cis, OVCAR-3, and A2780) but also to a healthy cell line (OSE). Although PAX was highly effective towards A2780cis cell line (11.9 ± 1.7 nM), it was not equally effective against another cisplatin-resistant cell line, OVCAR-3 (116.2 ± 2.4 nM). PAX obtained high values against A2780 and OSE cell lines as well (82 ± 10.5 nM and 70.3 ± 0.6 nM, respectively).

Table 5 IC₅₀ values of MTT Assay (μM: micromolar, nM: nanomolar, “-“: no data)

MTT IC ₅₀ results		Cell lines			
		A2780cis	OVCAR-3	A2780	OSE
Ligands	TSC ¹	80.5 ± 1.9 μM	210 ± 5.4 μM	-	-
	TSC ²	87.3 ± 1.1 μM	213 ± 6.1 μM	-	-
Drugs	Complex 1a	7.6 ± 2.0 μM	10.3 ± 0.7 μM	13.5 ± 1.3 μM	7.4 ± 2.2 μM
	Complex 1b	39.8 ± 1.1 μM	62.4 ± 9.4 μM	29.1 ± 1.5 μM	21.9 ± 8.6 μM
	Complex 2	46.2 ± 9.5 μM	60.9 ± 2.9 μM	24.2 ± 1.0 μM	33.6 ± 3.3 μM
	PAX	11.9 ± 1.7 nM	116.2 ± 2.4 nM	82 ± 10.5 nM	70.3 ± 0.6 nM
	CAR	41.7 ± 9.1 μM	234.5 ± 7.7 μM	70.7 ± 1.8 μM	41.6 ± 2.9 μM

Ligands of Ru agents were tested as a control to check whether the cause of cytotoxicity is from ligands or not. TSC¹ value for A2780cis was 80.5 ± 1.9 μM, and TSC² value for A2780cis was 87.3 ± 1.1 μM. TSC¹ value for OVCAR-3 was 210 ± 5.4 μM, and TSC² value for OVCAR-3 was 213 ± 6.1 μM. We showed that the cytotoxic effect was not from ligands of Ru agents. All Ru agents presented a higher anti-proliferative effect than ligands and generally lower cytotoxicity against healthy cell line (OSE) compared to carboplatin. Every compound exhibited distinctness in anti-proliferative activities.

Complex 1a was at least 3 to 5 times more effective than other Ru agents. Its IC₅₀ values were 7.6 ± 2.0 μM for A2780cis, 10.3 ± 0.7 μM for OVCAR-3, 13.5 ± 1.3 μM for A2780, 7.4 ± 2.2 μM for OSE cell line. On the other hand, complex 1b showed lower toxicity than complex 1a

such as $39.8 \pm 1.1 \mu\text{M}$ for A2780cis, $62.4 \pm 9.4 \mu\text{M}$ for OVCAR-3, $29.1 \pm 1.5 \mu\text{M}$ for A2780, $21.9 \pm 8.6 \mu\text{M}$ for OSE cell line. For complex 2 IC50 values were $46.2 \pm 9.5 \mu\text{M}$ for A2780cis, $60.9 \pm 2.9 \mu\text{M}$ for OVCAR-3, $24.2 \pm 1.0 \mu\text{M}$ for A2780, $33.6 \pm 3.3 \mu\text{M}$ for OSE cell line.

Carboplatin showed the lowest cytotoxicity in particular to OVCAR-3 cell line ($234.5 \pm 7.7 \mu\text{M}$) although it was highly toxic to OSE at most ($41.6 \pm 2.9 \mu\text{M}$). OVCAR-3 cell line was the most resistant cell type against all agents and failed to respond well while A2780cis cell line obtained better results than others.

4.2. Intra and Extra-Glucose Levels of Ovarian Cell Lines

Glucose metabolism was increased in cisplatin-resistant cells (A2780cis), while it was generally suppressed in cisplatin-sensitive cells (A2780). As can be seen in Figure 16, A2780cis, compared to a healthy cell line (OSE), obtained the highest intracellular glucose amount and the fastest glucose use in the cell. Compared to control, PAX and CAR treated OSE cells showed the highest rate of glucose use between 0 h to 24 h while Ru agents, particularly complex 1b and 2, showed the fastest glucose use between 24 h to 48 h (Figure 16A). Even though all cancer cell lines showed a high demand for glucose, cisplatin-resistant cell lines such as OVCAR-3 and A2780cis demonstrated the most excellent glucose uptake. All treatments in A2780 cells accelerated the glucose uptake compared to control cells (Figure 16B). Extracellular glucose was decreased more than half in all ovarian cancer cell lines, while OSE showed less glucose uptake compared to others in the first 24 h. Extracellular glucose in A2780cis cells dramatically decreased in PAX, CAR, and complex 2 treated cells compared to complex 1a and 1b treatment (Figure 16C). Complex 1a and 1b treatment in OVCAR-3 slowed the glucose uptake compared to control and other agents (Figure 16D).

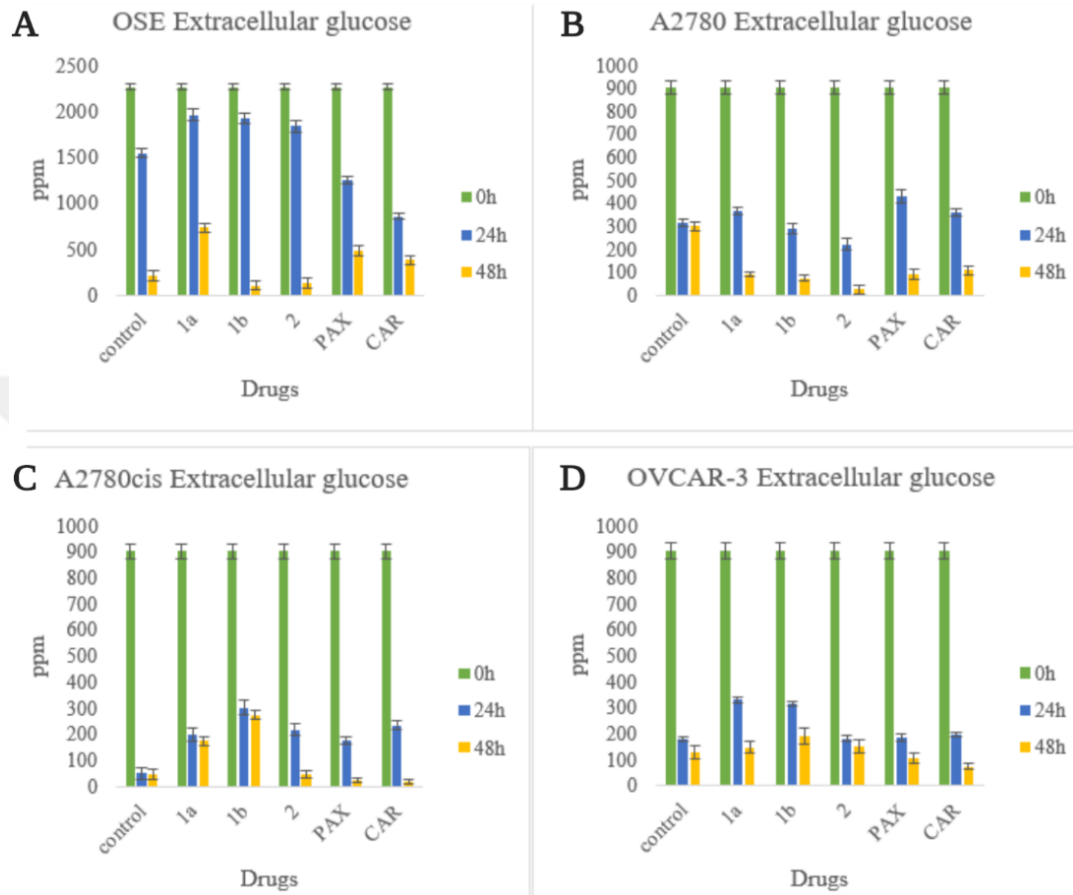


Figure 16 Change of extracellular glucose levels in ovarian cell lines (A: OSE cell line; B: A2780 cell line; C: A2780cis cell line; D: OVCAR-3 cell line).

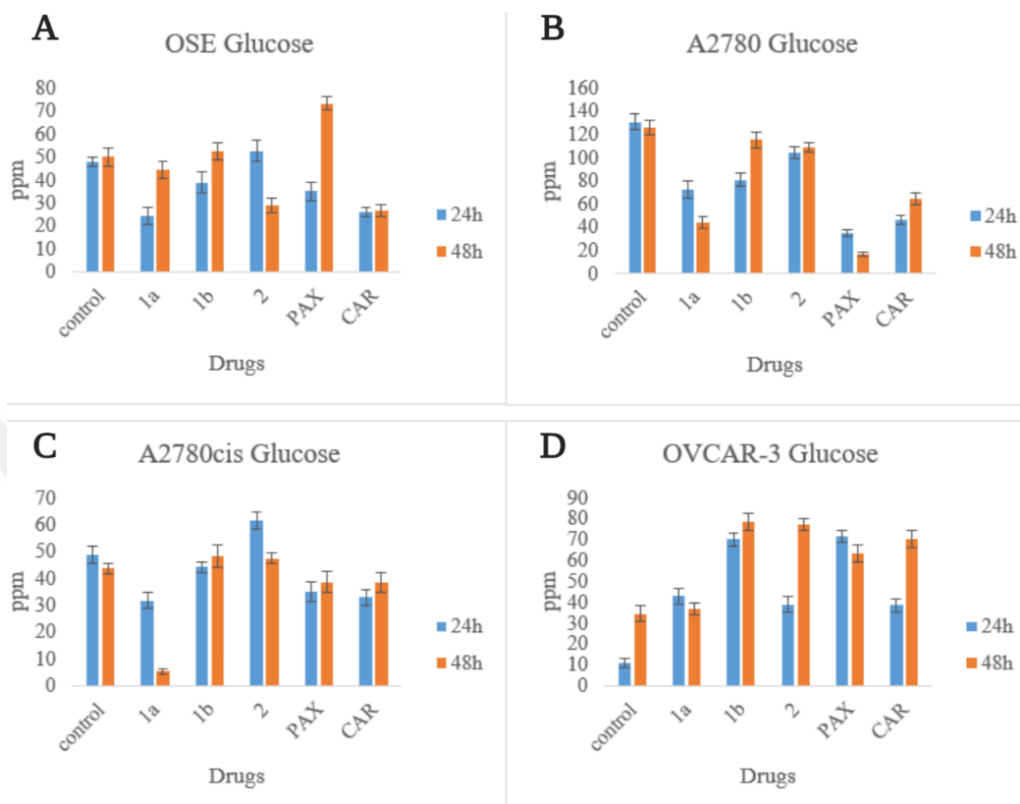


Figure 17 Change of intracellular glucose levels in ovarian cell lines (A: OSE cell line; B: A2780 cell line; C: A2780cis cell line; D: OVCAR-3 cell line).

Complex 1a, 1b, and PAX treated OSE cells showed an increase, while complex 2 showed a decrease in intracellular glucose between 24 h and 48 h compared to control (Figure 17A). Complex 1a and PAX treatment caused a decrease in intracellular glucose of A2780 cells compared to control. On the other hand, complex 1b, 2, and CAR treated A2780 cells showed an increased intracellular glucose value (Figure 17B). Glucose decreased dramatically in complex 1a treated A2780cis cells compare to others (Figure 17C). All treatments caused a high glucose uptake in OVCAR-3 cells, particularly complex 2 and CAR treatment, increased intracellular glucose levels two-fold higher between 24 h and 48 h (Figure 17D).

4.3. Intracellular TCA Cycle Metabolites of Ovarian Cell Lines

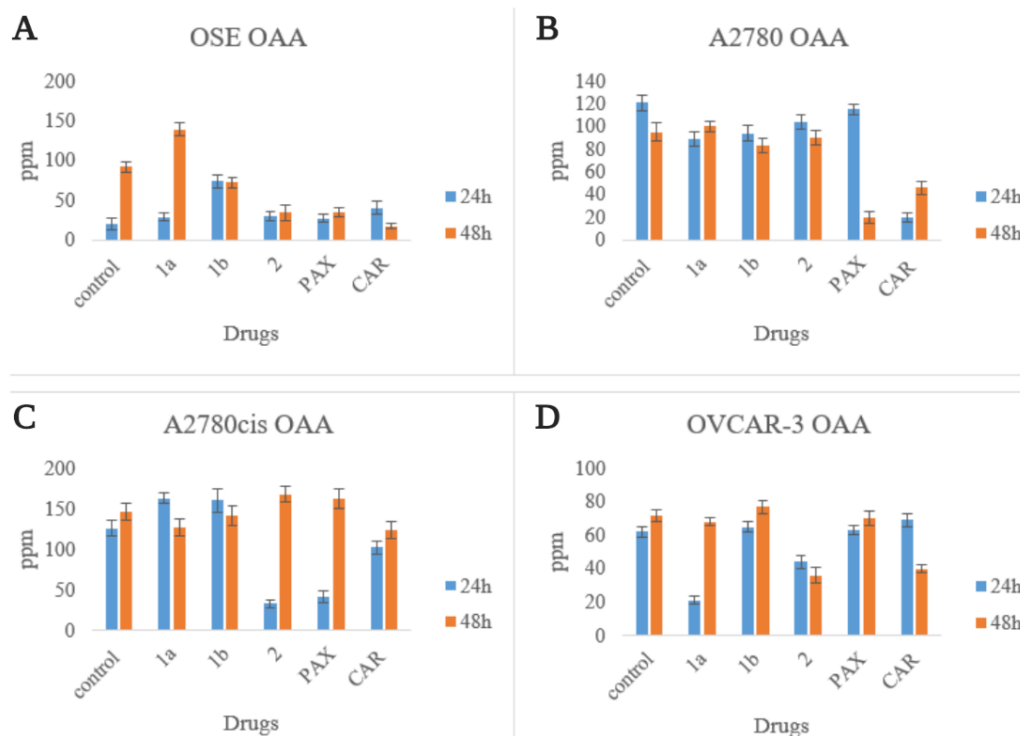


Figure 18 Change of OAA levels in ovarian cell lines (A: OSE cell line; B: A2780 cell line; C: A2780cis cell line; D: OVCAR-3 cell line).

Complex 1a increased OAA in OSE cells compared to control cells, while others were generally stable (Figure 18A). OAA reduced significantly in PAX treated A2780 cells between 24 h and 48 h while induced in CAR treated A2780 cells compared to control (Figure 18B). Although OAA amount was lower in CAR treated cells than in control cells, OAA increased significantly in complex 2, and PAX treated A2780cis cells while other agents slightly reduced it (Figure 18C). OVCAR-3 cells with complex 1a treatment caused a decrease in OAA values at 24 h, and CAR and complex 2 treatment also decreased OAA levels at 48 h compared to control (Figure 18D).

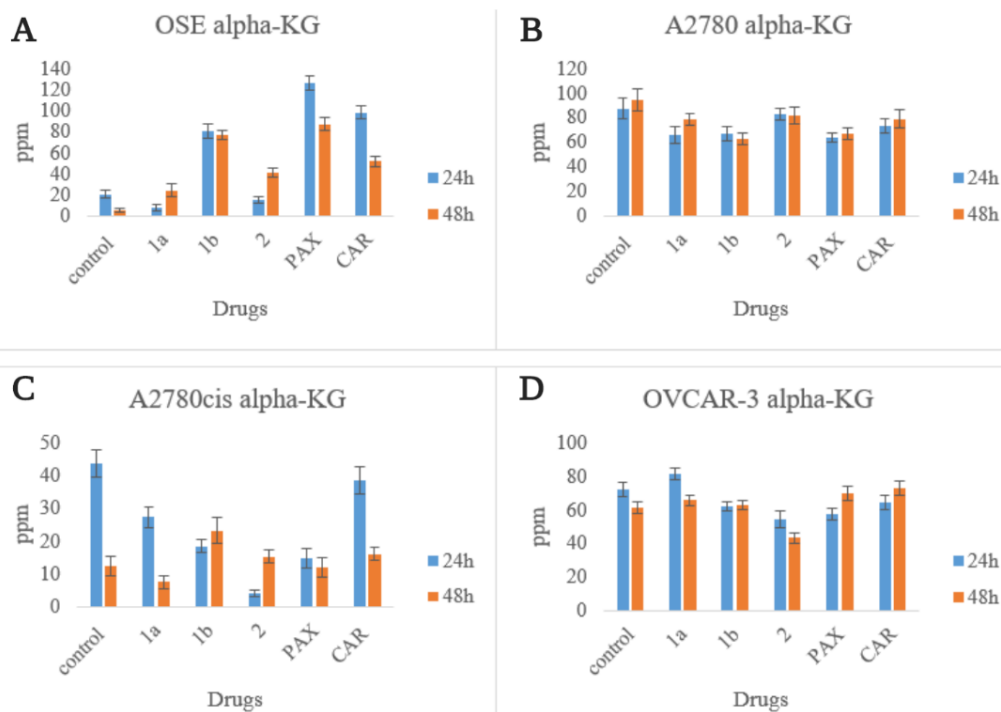


Figure 19 Change of alpha-KG levels in ovarian cell lines (A: OSE cell line; B: A2780 cell line; C: A2780cis cell line; D: OVCAR-3 cell line).

All treatments caused increased α KG levels in OSE cells compared to control, but a decrease in PAX and CAR treated OSE cells between 24 h and 48 h. On the contrary, complex 1a and 2 caused an increase compared to control (Figure 19A). Complex 1a treatment raised α KG levels of A2780 cells while other agents did not display much difference between 24 h and 48 h (Figure 19B). α KG levels in CAR and complex 1a treated A2780cis cells were significantly decreased while control was also decreased between 24 h and 48 h (Figure 19C). α KG levels in OVCAR-3 cells were stable mostly except complex 2 treated cells showed a decrease at 48 h (Figure 19D). In general, A2780cis cells had lower α KG levels than A2780 cells, and OVCAR-3 cells displayed stable α KG levels.

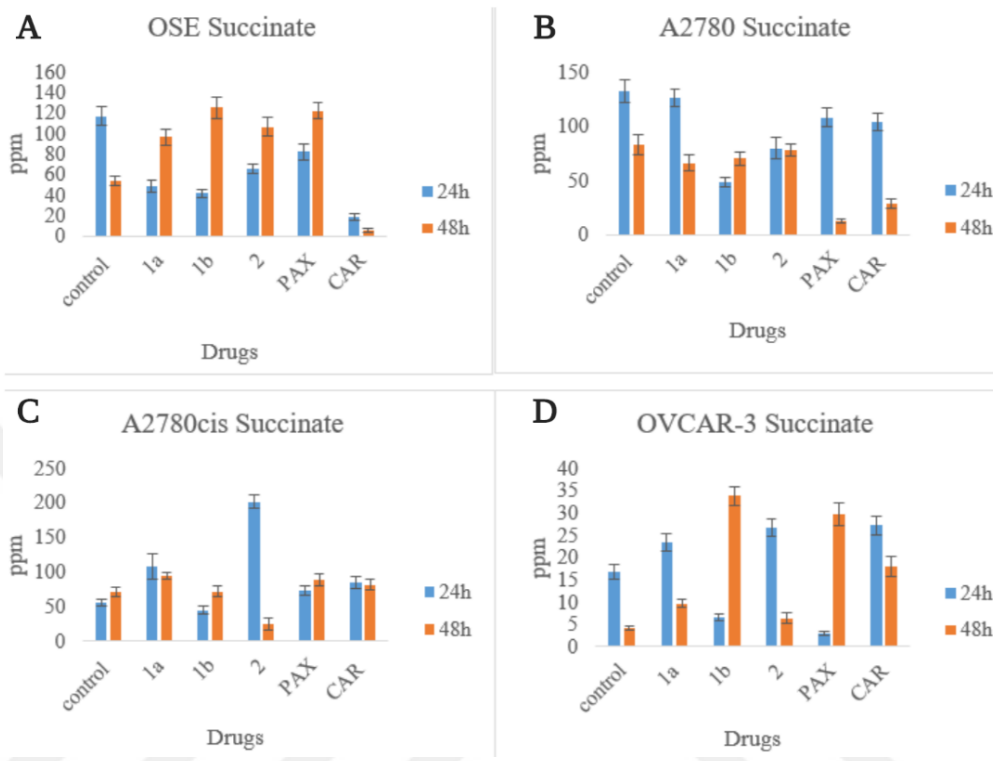


Figure 20 Change of succinate levels in ovarian cell lines (A: OSE cell line; B: A2780 cell line; C: A2780cis cell line; D: OVCAR-3 cell line).

Ru agents and PAX increased the succinate levels in OSE cells significantly while it was decreased in control cells. Also, CAR showed the lowest succinate levels in OSE cell line (Figure 20A). Complex 1a, PAX, and CAR treated A2780 cells displayed a significant reduction in succinate between 24 h and 48 h. On the other hand, complex 2 and 1b also lowered the succinate level at 24 h compared to control (Figure 20B). Complex 2 increased succinate levels in A2780cis cells at 24 h but then decreased dramatically at 48 h. Although other agents did not affect the succinate levels as much as the complex 2, complex 1a also increased while complex 1b decreased it at 24 h (Figure 20C). PAX and complex 1b treatment first decreased succinate levels at 24 h then increased in OVCAR-3 cells dramatically between 24 h and 48 h while complex 1a, 2, and CAR first increased it at 24 h then decreased it between 24 h and 48 h (Figure 20D). In general, succinate in cisplatin-resistant cells (OVCAR-3 and A2780cis) decreased more compared to cisplatin-sensitive cells (OSE and A2780).

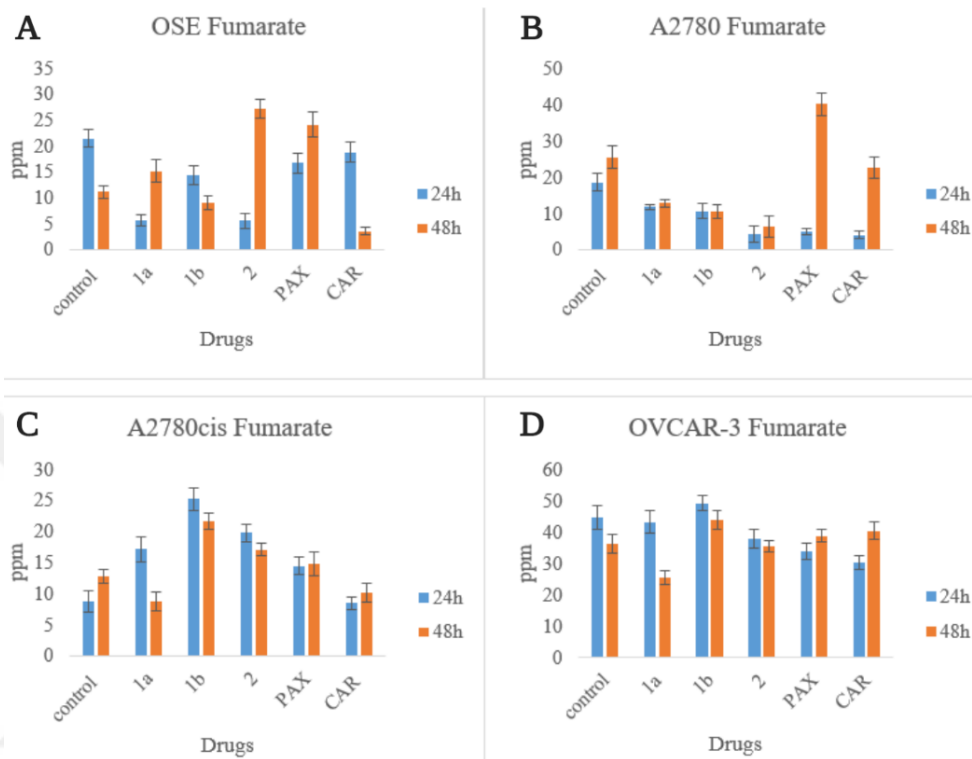


Figure 21 Change of fumarate levels in ovarian cell lines (A: OSE cell line; B: A2780 cell line; C: A2780cis cell line; D: OVCAR-3 cell line).

Fumarate levels in OSE reduced after complex 2, 1a, and 1b treatments at 24 h. However, complex 1a, 2, and PAX treated OSE cells increased fumarate levels at 48 h while CAR decreased it (Figure 21A). All Ru agents decreased fumarate levels in A2780 cells. PAX and CAR caused a decrease at 24 h but then increased notably at 48 h compared to control (Figure 21B). Complex 1a, 1b, 2 and PAX elevated fumarate levels in A2780cis cells at 24 h. Complex 1a and CAR decreased it at 48 h while complex 1b, 2, and PAX increased it compared to control (Figure 21C). Complex 2, PAX, and CAR lowered fumarate levels in OVCAR-3 cells at 24 h. Complex 1b, PAX, and CAR increased it at 48 h while complex 1a decreased it (Figure 21D). Generally, OVCAR-3 cells displayed mostly stable but significantly high fumarate levels compare to others.

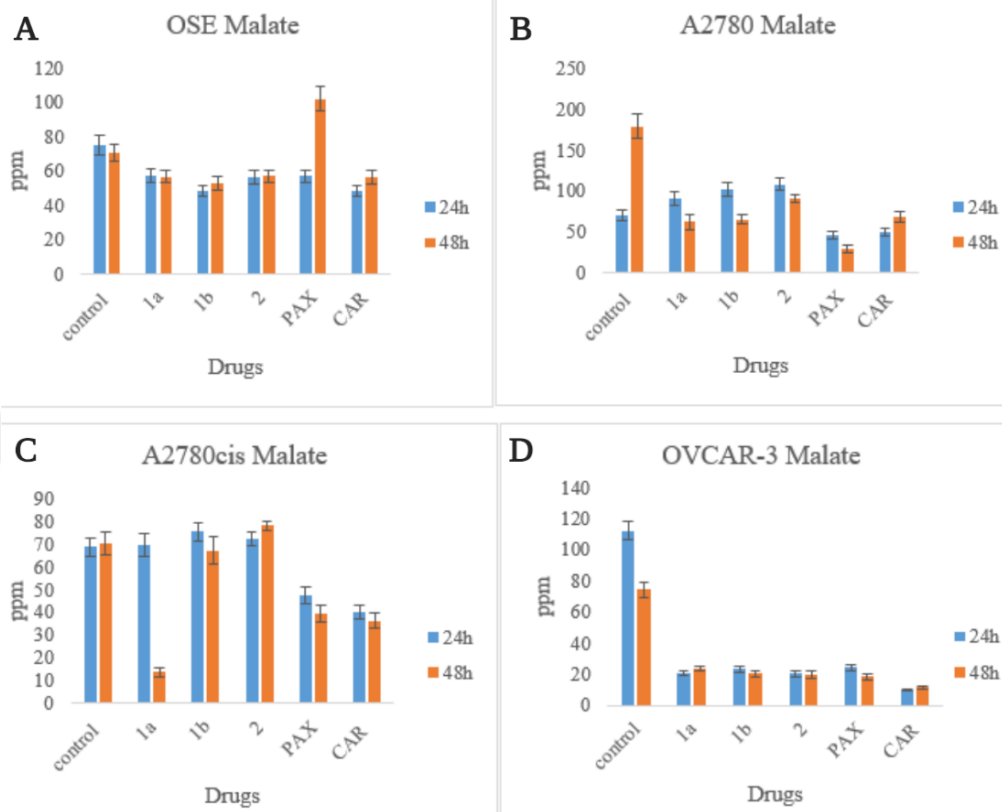


Figure 22 Change of malate levels in ovarian cell lines (A: OSE cell line; B: A2780 cell line; C: A2780cis cell line; D: OVCAR-3 cell line).

Ru agents and CAR treatments reduced malate levels in OSE cells compared to control, while PAX treatment lowered it significantly at 48 h (Figure 22A). PAX and CAR decreased malate levels in A2780 cells compared to control. Ru agents increased it at 24 h but then decreased it at 48 h compared to control cells (Figure 22B). Complex 1a, PAX, and CAR treatment decreased malate levels substantially in A2780cis cells, while others did not display any specific change (Figure 22C). All treatments, especially CAR, caused a dramatic decrease in malate levels of OVCAR-3 cells to compare to control (Figure 22D).

4.4. Lactic Acid Production in Ovarian Cell Lines

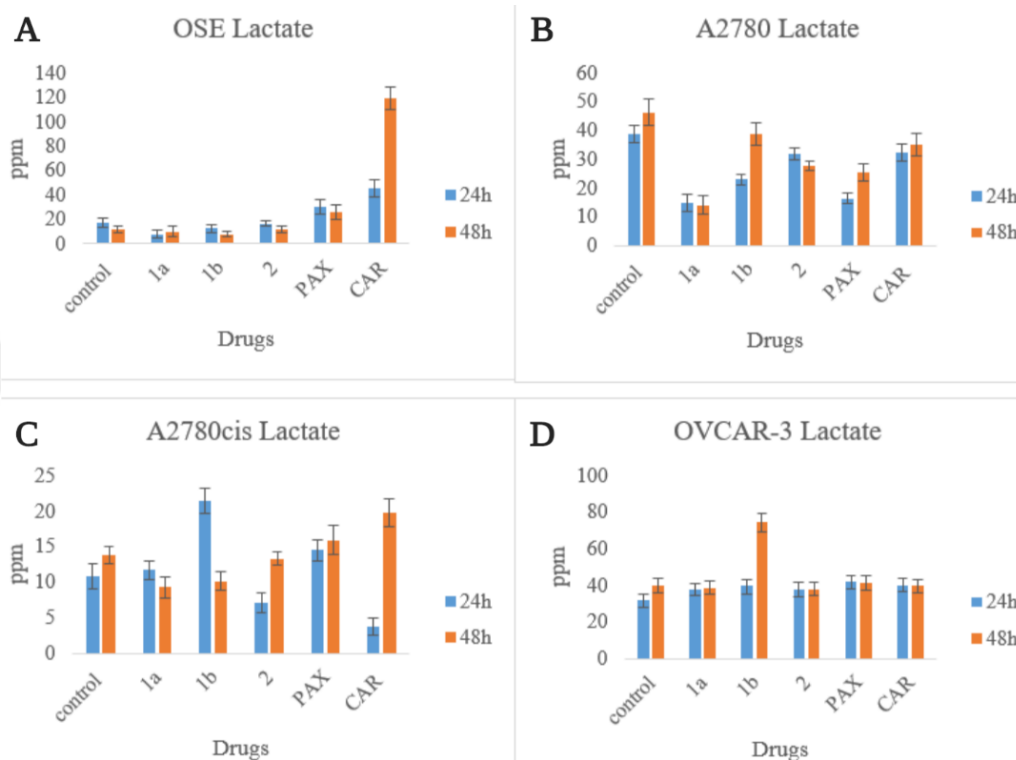


Figure 23 Change of lactate levels in ovarian cell lines (A: OSE cell line; B: A2780 cell line; C: A2780cis cell line; D: OVCAR-3 cell line).

Although other agents did not change lactate levels in OSE cells, PAX and CAR treatment considerably increased lactate levels in OSE cells compared to control (Figure 23A). All agents decreased lactate in A2780 cells, but especially complex 1a, 1b, and PAX reduced significantly at 24 h, but then PAX treated cells increased at 48h (Figure 23B). Complex 1b caused an increase in A2780cis cells at 24 h but reduced it at 48 h. On the other way, CAR and complex 2 treatment caused a decrease at 24 h but then increased it at 48 h. Complex 1a and PAX did not make any significant change in lactate levels of A2780cis cells compared to control cells (Figure 23C). Complex 1b increased lactate levels in OVCAR-3 cells at 48 h, while others did not change much. Generally, OVCAR-3 displayed stable lactate levels (Figure 23D).

5. DISCUSSION

Cell viability assessments showed that complex 1a has promising anti-proliferative attributes against cisplatin-resistant ovarian cancer cell lines, especially against OVCAR-3 cell line, which was the most resistant type among all. However, it was highly toxic to a healthy cell line (OSE), as well. Complex 2 displayed similar cytotoxicity levels with complex 1b, but its toxicity was lower against OSE cells, and also complex 2 was more efficient against A2780 cell line than complex 1b. Among three Ru agents that were used complex 1a was at least 3 to 5 times more effective than other ruthenium agents.

The cytotoxicity of Ru agents generally decreased at 48 h compared to 24 h values. As the cellular uptake experiments (Tavsan, Yaman, Subasi, & Ayar Kayalı, 2018) confirmed that Ru agents that were used were more efficient until 12 h because after that, it was started to throw out of the cells. Therefore complex 1b and 2 showed lower cytotoxicity in cells, and thus, cell proliferation was continued at 48 h. It was led to more glucose uptake and consumption in the cells. However, complex 1a showed an opposite aspect to other Ru agents; it was still highly toxic even after 24 h.

PAX was severely toxic to all cell lines, even with the concentration of nanomolar grade. Additionally, its toxicity gets higher at 48 h even with 5 nM dose. Although it was efficient against A2780cis (cisplatin-resistant ovarian cell line), it failed to demonstrate the same success towards OVCAR-3, which is another cisplatin-resistant cell line. Moreover, it was also harmful to a healthy cell line (OSE). However, CAR failed to kill OVCAR-3 cells while it was highly toxic towards OSE and A2780cis cells as much as complex 2.

Increased glycolysis is a crucial hallmark during cancer progression (Warburg, 1956b; Mathupala, Rempel, & Pedersen, 2001; Mathupala, Ko, & Pedersen, 2006). Increases in glucose and glutamine uptake are crucial for highly proliferating cells because they regularly require precursors for the synthesis of fatty acids and nucleic acids (Desideri, Vegliante, & Ciriolo, 2015).

It was reported that cisplatin-resistant ovarian cancer cells showed a high glucose demand (Catanzaro et al., 2015). HPLC analysis proved that the cisplatin-resistant cell line (A2780cis) possesses accelerated glucose metabolism than cisplatin-sensitive cells (A2780). Furthermore,

extracellular glucose levels were decreased more than half in all ovarian cancer cell lines, while OSE cell line exhibited reduced glucose uptake at 24 h. Extracellular glucose dramatically reduced in PAX, CAR, and complex 2 treated A2780cis cells compared to complex 1a and 1b. These conclusions may indicate a weakened glucose uptake in OVCAR-3 cells effected by complex 1a and 1b treatments.

Complex 1a and PAX treatments caused a decrease in intracellular glucose of A2780 cells compared to control. However, complex 1b, 2, and CAR applied A2780 cells exhibited increased intracellular glucose levels. Glucose decreased dramatically in complex 1a applied A2780cis cells compare to others. All agents caused high glucose uptake in OVCAR-3 cells; however, complex 2 and CAR treatments raised the intracellular glucose levels two-fold higher at 48 h compared to 24 h.

Each agent displayed a unique effect in cellular metabolism as each cell line responded differently to the agents. It is accepted that the TCA cycle oncometabolite accumulation triggers and supports the progression of cancer (Menendez, Alarcon, & Joven, 2014; Nam et al., 2014). Another study found that aberrations of the TCA cycle are related to various kinds of cancers (Montal et al., 2015; Pavlova & Thompson, 2016; Yuan et al., 2016). The mitochondrial aberrations lead to metabolic alteration in tumor cells and raise the rate of glycolysis, which supports cell survival and proliferation in turn (Frezza & Gottlieb, 2009; Gaude & Frezza, 2014).

Glycogen synthesis is commonly observed in cancer cells under hypoxic circumstances (Iida et al., 2012; Uekuri et al., 2013). Additionally, it is also related to resistance against chemotherapeutic drugs (Iida et al., 2012). Complex 1a applied A2780cis cells demonstrated accelerated glucose consumption but reduced TCA cycle metabolites in general. These results may imply the usage of glucose in glycogen synthesis or nucleotide synthesis through the pentose phosphate pathway or other pathways.

Glucose transporters are overexpressed to compensate for the increased glycolysis, and thus, glucose levels are elevated in cancer cells (Macheda, Rogers, & Best, 2005; Pelicano, Martin, Xu, & Huang, 2006). Interestingly, in cancer cells, lactate production via aerobic glycolysis does not shut down the OXPHOS pathway completely (Israelsen et al., 2013). In agreement with this,

A2780cis cell line with complex 2 treatment showed elevated lactic acid production as well as the α KG, OAA, and fumarate levels. These results may indicate a proof for a metabolic shift in cellular metabolism toward anaerobic respiration but also with accelerated TCA cycle. Furthermore, complex 1a applied A2780 cells showed lowered lactate levels with increased α KG and OAA levels. It may refer to a suppression of the lactic acid production in A2780 cell line with the acceleration of the TCA cycle.

Complex 1b applied OVCAR-3, and A2780 cells showed similar results as increased glucose uptake and lactate levels. These findings may imply a shift in cellular metabolism toward anaerobic respiration. In contrast to this, A2780cis cells displayed elevated TCA cycle intermediates with the same treatment, which might indicate an increased OXPHOS in cells.

In OVCAR-3 cells, PAX application caused an increase in TCA cycle intermediates and intracellular glucose levels. It may infer an accelerated OXPHOS in ovarian cancer cell lines by PAX treatment. Additionally, CAR treatment suppressed the TCA cycle in almost all cell lines and elevated the lactate levels in OSE cells, which may indicate an increase in anaerobic respiration along with suppressed OXPHOS.

One of the crucial intermediates of the TCA cycle is α KG because of its role in glutaminolysis, which is an energy-generating pathway in cellular metabolism. It is accepted that cancer cells overexpress the enzymes that are necessary for the conversion of glutamate from glutamine to supply additional energy sources under circumstances of glucose deprivation. Glutamate is used to create α KG, which can replenish the TCA cycle to synthesize ATP or amino acid (Yuneva et al., 2007; Weinberg et al., 2010; Altman, Stine, & Dang, 2016). Moreover, OAA can be overproduced to use in nucleotide synthesis via conversion to aspartate (Son et al., 2013).

α KG levels expected to be reduced in IDH-mutant cancer cells as a result of increased α KG conversion to 2-Hydroxyglutarate (2HG). However, it is rather elevated as a result of mitochondrial biosynthesis to compensate for the absent α KG (van Lith et al., 2014). In agreement with this study, α KG levels of OVCAR-3 cell line were generally stable but high. After treatment with agents, OSE cells showed a similar profile to OVCAR-3 cells and raised cellular α KG levels dramatically

compared to control. These results may suggest that OSE cells reprogrammed their metabolism to adapt to certain conditions like chemotherapy and imitated a profile alike to a resistant cell.

Another critical intermediate of the TCA cycle is succinate, which owns a significant part in producing ATP. Additionally, it is a fundamental modulator of the hypoxic response and has a significant role in tumorigenesis. SDH is responsible for transforming succinate to fumarate (King, Selak, & Gottlieb, 2006). Especially, mutations in the gene encoding SDH were involved in numerous cancers. Therefore, SDH is now considered a tumor suppressor (Bardella, Pollard, & Tomlinson, 2011). Also, SDH mutations cause an elevation in succinate levels and lead to HIF-1 α (Hypoxia-inducible factor 1- α) stabilization, thus supporting tumor growth (Guzy et al., 2008).

Some studies reported that under nutrient deprivation or hypoxic conditions, cancer cells rely on fumarate respiration (Tomitsuka, Kita, & Esumi, 2010; Sakai et al., 2012). Fumarate respiration occurs by converting fumarate to succinate through the reverse reaction of SDH, which further generates ATP (Kita & Takamiya, 2002). Succinate, the outcome of fumarate respiration was observed remarkably high in tumor cells (Hirayama et al., 2009). In agreement with this, complex 1b treated OVCAR-3 and OSE cells, and complex 1a treated A2780cis cells exhibited high succinate levels.

Loss-of-function mutations in FH and SDH, increase the amount of fumarate and succinate, respectively (Pollard *et al.*, 2005), and gain-of-function mutations in IDH elevate D-2-hydroxyglutarate (D-2HG) levels (Ward et al., 2010; Dang et al., 2010). Another study proved that fumarate acts as an oncometabolite, which means metabolites that contribute to oncogenic processes (Sciacovelli & Frezza, 2016), and leading to over-expression of EMT-driving transcription factors. Complex 1a applied OSE cells demonstrated increased TCA cycle intermediates. However, they also showed lowered fumarate levels, which might be due to fumarate respiration in cells.

Complex 1a applied OVCAR-3 cells demonstrated decreased fumarate and OAA levels while OVCAR-3 cells with PAX treatment showed increased succinate, α KG, and fumarate levels. These results may imply a change in the energy metabolism of OVCAR-3 cells from fumarate respiration to OXPHOS respect to the alteration of treatment of complex 1a to PAX. All Ru agents

decreased fumarate levels in A2780 cells while Ru agents and PAX increased fumarate levels in A2780cis cells at 24 h.

In conclusion, we demonstrated the cytotoxic effects of Ruthenium complexes as well as carboplatin and paclitaxel drugs on the energy metabolism in ovarian cell lines. We found that Ruthenium complexes are promising candidates, and with further improvements, their potential might increase the efficiency of treatment against ovarian cancer. These results proved that each chemotherapeutic agent caused a characteristic metabolic response in ovarian cell lines. This study may lead to further research for better therapeutics against cancer and improve the knowledge of the effects of chemotherapeutics on the energy metabolism of ovarian cells.

6. CONCLUSION AND RECOMMENDATIONS

Every cancer cell displays individual metabolic phenotypes of energy for two main reasons. First, it has been acknowledged that cancer is a heterogeneous disease, and it possesses genetic heterogeneity, which leads to metabolic heterogeneity in the cell (Marusyk & Polyak, 2010). Even in the same type of cancer, its constituent cell subtypes also exhibit characteristic variations in metabolic phenotype from one cell to another (Feron, 2009). Second, cancer cells constantly reprogram to adapt to challenging environmental alterations and nutrients deprivations. Therefore we understand that balances between glycolysis and OXPHOS or other pathways to produce ATP are continuously changing in cancer cells. Our results demonstrated that each cell line displayed a distinct metabolic response to the chemotherapeutic agents. Ruthenium complexes have promising attributes and can be highly beneficial in cancer treatment with further improvements. It is an essential step to unveil the effects of chemotherapeutic agents on cellular metabolism in order to discover new novel therapeutic approaches against cancer.

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8. APPENDIX

8.1. Curriculum Vitae

Gölnur Çırak

Birth Year:	1996
Address:	İzmir Uluslararası Biyotıp ve Genom Enstitüsü (İBG) Dokuz Eylül Üniversitesi Sağlık Yerleşkesi Balçova 35340 İzmir
e-mail:	gulnircirak@gmail.com

EDUCATION

Country	University	Faculty/Institute	Field of Study	Degree	Graduation Year
TR	Dokuz Eylül University	Izmir International Biomedicine and Genome Institute	Molecular Biology and Genetics	MSc	2019
TR	Istanbul University	Faculty of Science	Molecular Biology and Genetics	BA	2017

ACADEMIC EXPERIENCES

University	Country	City	Department	Type	Date
Izmir Institute of Technology	TR	Izmir	Molecular Biology and Genetics	Internship	July-August 2016
Ege University	TR	Izmir	Bioengineering, BILTEM	Internship	January-February 2015
Ege University	TR	Izmir	Faculty of Medicine, Department of Medical Biology	Internship	July 2014

RESEARCH FIELDS

Molecular Cancer Biology, Energy Metabolism, Retrotransposons, Gene editing
